

QUESTION 10 (Start on a new page.)

The balanced chemical equation for the EXTRACTION of gold from its ore is as follows:



- 10.1 State ONE disadvantage of using cyanide (CN^-) in the extraction of gold. (1)
- 10.2 Will the final solution of the extraction process be ACIDIC or BASIC (ALKALINE)? Give a reason for the answer. (2)
- 10.3 Determine the oxidation number of gold in NaAu(CN)_2 . (1)
- 10.4 Write down the FORMULA of the reducing agent in the reaction above. (1)

Zinc powder is now used to PRECIPITATE the gold.

The balanced equation for the reaction is:



- 10.5 Does zinc undergo OXIDATION or REDUCTION during the precipitation reaction? (1)
- 10.6 Write down a half-reaction to support the answer to QUESTION 10.5. (2)
- 10.7 Calculate the percentage of gold in NaAu(CN)_2 . (2)
- [10]**

TOTAL: 150

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Table of Contents Physical Sciences P2 Caps Grade 1dbe November 2014

1. Understanding the eBook Physical Sciences P2 Caps Grade 1dbe November 2014
 - The Rise of Digital Reading Physical Sciences P2 Caps Grade 1dbe November 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Sciences P2 Caps Grade 1dbe November 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Sciences P2 Caps Grade 1dbe November 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Sciences P2 Caps Grade 1dbe November 2014
 - Personalized Recommendations
 - Physical Sciences P2 Caps Grade 1dbe November 2014 User Reviews and Ratings
 - Physical Sciences P2 Caps Grade 1dbe November 2014 and Bestseller Lists
5. Accessing Physical Sciences P2 Caps Grade 1dbe November 2014 Free and Paid eBooks
 - Physical Sciences P2 Caps Grade 1dbe November 2014 Public Domain eBooks
 - Physical Sciences P2 Caps Grade 1dbe November 2014 eBook Subscription Services
 - Physical Sciences P2 Caps Grade 1dbe November 2014 Budget-Friendly Options

6. Navigating Physical Sciences P2 Caps Grade 1dbe November 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Sciences P2 Caps Grade 1dbe November 2014 Compatibility with Devices
 - Physical Sciences P2 Caps Grade 1dbe November 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Sciences P2 Caps Grade 1dbe November 2014
 - Highlighting and Note-Taking Physical Sciences P2 Caps Grade 1dbe November 2014
 - Interactive Elements Physical Sciences P2 Caps Grade 1dbe November 2014
8. Staying Engaged with Physical Sciences P2 Caps Grade 1dbe November 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Sciences P2 Caps Grade 1dbe November 2014
9. Balancing eBooks and Physical Books Physical Sciences P2 Caps Grade 1dbe November 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Sciences P2 Caps Grade 1dbe November 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Sciences P2 Caps Grade 1dbe November 2014
 - Setting Reading Goals Physical Sciences P2 Caps Grade 1dbe November 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Sciences P2 Caps Grade 1dbe November 2014
 - Fact-Checking eBook Content of Physical Sciences P2 Caps Grade 1dbe November 2014
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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