

Lesson Quiz

Observing Chemical Change

Write the letter of the correct answer on the line at the left.

1. ____ Which of the following is true about chemical reactions?
A They are accompanied by changes in energy.
B They form new substances with new properties.
C both A and B
D neither A nor B
2. ____ In an endothermic reaction, energy is
A absorbed
B released
C converted to mass
D synthesized
3. ____ Which of the following is NOT a physical property?
A melting point
B state of matter
C density
D flammability
4. ____ Substances formed as a result of a chemical reaction are called
A catalysts
B precipitates
C products
D reactants

If the statement is true, write *true*. If the statement is false, change the underlined word or words to make the statement true.

5. _____ In an exothermic reaction, products have more energy than reactants.
6. _____ Water boils at 100°C. This is an example of a chemical property.
7. _____ Substances that enter into a chemical reaction are called products.
8. _____ The ability to react with oxygen is an example of a chemical property.
9. _____ Another name for a chemical change is a chemical bond.
10. _____ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

Observing Chemical Change Answers Pearson

**Pearson/Prentice Hall, Michael J.
Padilla**



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Prentice Hall Science Explorer: Chemical Interactions Pearson/Prentice Hall, Michael J. Padilla, 2004-10

Pearson Edexcel A Level Chemistry (Year 1 and Year 2) Andrew Hunt, Graham Curtis, Graham Hill, 2019-07-15

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Focus on Physical Science California Edition Michael J. Padilla, 2007

FCS physical science L2, 2007

Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking

Constructing Science in Elementary Classrooms Norman G. Lederman, Judith S. Lederman, Randy L. Bell, 2004 Solidly based on the National Science Education Standards and Benchmarks for Science Literacy this new elementary science methods text immerses students into the context of classroom instruction through the authors unique approach using The Teaching Cycle The text is divided into three major sections or clusters of chapters Goals of Science Instruction Setting the Stage and The Teaching Cycle The first two sections provide the theoretical and practical foundations for instruction while the third section provides content Section I presents an overall view of science as a way of knowing and eventually develops an argument for why science should be included in the curriculum at all Section II emphasizes the importance of connecting lessons and avoiding the tendency to present individual science lessons in isolation The major content chapters comprising Section III The Teaching Cycle Life Environmental Science Physical Science Earth and Space Science each incorporate the traditional topics of methods courses e g demonstrations laboratories classroom management assessment developmental psychology etc concepts and themes common to national reforms

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