

3. A student wants to remove an acid impurity from an organic liquid.
What should the student do?

A. Add $\text{Na}_2\text{CO}_3(\text{aq})$
B. Reflux the mixture
C. Add Br_2
D. Add MgSO_4

Your answer ☐

[1]

4. For which purpose is distillation used?
- A. to allow a liquid to boil without the loss of vapour
B. to purify a liquid product
C. to remove an involatile impurity
D. to allow further reaction without the loss of product

Your answer ☐

[1]

5. The reaction below produces a chloroalkane.



Which row shows the steps to purify the liquid product in the correct order?

A	Use a separating funnel	Distil	Dry	Remove unreacted HCl
B	Remove unreacted HCl	Use a separating funnel	Dry	Distil
C	Remove unreacted HCl	Use a separating funnel	Distil	Dry
D	Use a separating funnel	Remove unreacted HCl	Distil	Dry

Your answer ☐

[1]

Ocr Chemistry As Mark Scheme 23 May 2014

D Siedentop



Ocr Chemistry As Mark Scheme 23 May 2014:

This book delves into Ocr Chemistry As Mark Scheme 23 May 2014. Ocr Chemistry As Mark Scheme 23 May 2014 is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Ocr Chemistry As Mark Scheme 23 May 2014, encompassing both the fundamentals and more intricate discussions.

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- Chapter 4: Ocr Chemistry As Mark Scheme 23 May 2014 in Specific Contexts
- Chapter 5: Conclusion

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