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Julia Schneider



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web choose 1 answer the solution becomes negatively charged due to the majority Cl^- ions a the solution becomes negatively charged due to the majority Cl^- ions the solution becomes positively charged due to the stronger Mg^{2+} ions b the solution

becomes positively charged due to the stronger Mg^{2+} ions the solution remains neutrally

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web answer d somewhere in region d justification with the test charge and c 1 being negative there is a repulsive force on the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb s law $\frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$ 1 $\frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$ 2 we know that the force

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