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B.Sc. (Hons.)/III/NS

F

CHEMISTRY—Paper XVII

(Physical Chemistry—III)

Time : 3 Hours

Maximum Marks : 38

(Write your Roll No. on the top immediately on receipt of this question paper.)

Attempt six questions in all.

Q. No. 1 is compulsory. At least one question
is to be attempted from each Section.

1. Explain briefly any *four* of the following : $4 \times 2 = 8$
- (a) Maximum order of a reaction is restricted to three only.
 - (b) A photochemical reaction with quantum efficiency of one shows no temperature dependence.
 - (c) Adsorption of gases on the surface of adsorbent is not monolayer at high pressure and low temperature.
 - (d) Why can we not use a voltmeter for determining the electromotive force of a galvanic cell ?
 - (e) A first order reaction never completes.

P.T.O.

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