

T.Y.B.Sc. Semester 5 Mock Exam Question Paper
Chemistry Paper I – Physical Chemistry
Multiple Choice Questions

- Q1 Vapour pressure of a liquid solventwhen a non volatile solute is dissolve in it. 1M
- a) Increase
 - b) Decrease
 - c) no change
 - d) Expand
- Q2 Raoult's law is expressed by.....
- a) $\frac{P^o - P}{P^o} = X_2$
 - b) $\frac{P - P^o}{P^o} = X_2$
 - c) $\frac{P - P^o}{P} = X_2$
 - d) $\frac{P - P^o}{P^o} = X$
- Q3 is used in Rast method because it has a high value of K_f .
- a) Camphor
 - b) Bi phenyl
 - c) Naphthalene
 - d) Sodium chloride
- Q4 Boiling point elevation of the solution ΔT_b is given as.....
- a) $\Delta T_b = T - T_o$
 - b) $\Delta T_b = T_o - T$
 - c) $\Delta T_b = T + T_o$
 - d) $\Delta T_b = T \times T_o$
- Q5 The unit for activation of energy is
- a) kJ mol^{-1}
 - b) kJ K^{-1}
 - c) $\text{kJ K}^{-1} \text{mol}^{-1}$
 - d) kJ R^{-1}

Physical Chemistry Multiple Choice Questions

Peter William Atkins, Julio De Paula



Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25

Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students

Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas, 1970-01-01 Physical Chemistry M.M.L. Oblitas, 1977 Physical Chemistry M.M.L. Oblitas, 1971 *The Pearson Guide to Physical Chemistry for the IIT JEE* Singhal, *A Textbook of Physical Chemistry, 6th Edition* Sharma K.K. & Sharma L.K., 2016 *A Textbook of Physical Chemistry* **Elements of Physical Chemistry** Peter Atkins, Julio de Paula, 2013 *Elements of Physical Chemistry* has been carefully crafted to help students increase their confidence when using physics and mathematics to answer fundamental questions about the structure of molecules how chemical reactions take place and why materials behave the way they do *Class 9 Chemistry MCQ (Multiple Choice Questions)* Arshad Iqbal, *The Class 9 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF* 9th Grade Chemistry MCQ PDF Download Quiz Questions Chapter 1 8 Practice Tests with Answer Key Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 9 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 9 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 9 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 9 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters Chemical reactivity electrochemistry fundamentals of chemistry periodic table and periodicity physical states of matter solutions structure of atoms structure of molecules tests for school and college revision guide Class 9 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 9 Chemistry MCQs Chapter 1 8 PDF includes high school question papers to review practice tests for exams Class 9 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam 9th Grade Chemistry Mock Tests Chapter 1 8 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Chemical Reactivity MCQ Chapter 2 Electrochemistry MCQ Chapter 3 Fundamentals of Chemistry MCQ Chapter 4 Periodic Table and Periodicity MCQ Chapter 5 Physical States of Matter MCQ Chapter 6 Solutions MCQ Chapter 7 Structure of Atoms MCQ Chapter 8 Structure of Molecules MCQ The Chemical Reactivity MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Metals and non metals The Electrochemistry MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Corrosion and prevention

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AQA A-level Year 2 Chemistry Student Guide: Physical chemistry 2 Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions Develops independent learning and research skills Provides the content for generating individual revision notes

Working Method Approach for Introductory Physical Chemistry Calculations Brian J Hathaway, Clair Murphy, Brian Murphy, 2019-05-16 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical

approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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