

Periodic Table of the Elements



11.01

H

Hydrogen

3.04

Li

Lithium

6.94

Be

Beryllium

22.99

Na

Sodium

24.31

Mg

Magnesium

39.10

K

Potassium

40.08

Ca

Calcium

88.62

Rb

Rubidium

87.62

Cs

Cesium

226.03

Fr

Francium

14

28.09

Si

Silicon

atomic number

atomic weight

symbol

name

black

solid

blue

liquid

red

gas

white

synthetically prepared

near ambient temp

alkali metals

alkaline earth metals

transitional metals

other metals

nonmetals

noble gases

2

4.00

He

Helium

5

6.94

B

Boron

6

12.01

C

Carbon

7

14.01

N

Nitrogen

8

16.00

O

Oxygen

9

18.99

F

Fluorine

10

20.18

Ne

Neon

13

26.98

Al

Aluminum

14

28.09

Si

Silicon

15

30.97

P

Phosphorus

16

32.06

S

Sulfur

17

35.45

Cl

Chlorine

18

39.96

Ar

Argon

31

69.72

Ga

Gallium

32

72.64

Ge

Germanium

33

74.92

As

Arsenic

34

78.96

Se

Selenium

35

79.90

Br

Bromine

36

83.90

Kr

Krypton

49

114.82

In

Indium

50

118.71

Sn

Tin

51

121.76

Sb

Antimony

52

127.60

Te

Tellurium

53

126.91

I

Iodine

54

131.29

Xe

Xenon

61

151.96

Tl

Thallium

62

157.25

Pb

Lead

63

158.91

Bi

Bismuth

64

158.91

Po

Polonium

65

158.91

At

Astatine

66

158.91

Rn

Radon

81

200.59

Tl

Thallium

82

200.59

Pb

Lead

83

200.59

Bi

Bismuth

84

200.59

Po

Polonium

85

200.59

At

Astatine

86

200.59

Rn

Radon

113

204.38

(113)

114

209

(114)

115

209

(115)

116

209

(116)

117

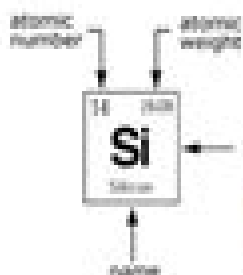
209

(117)

118

209

(118)



black solid
 blue liquid
 red gas
 white synthetically prepared
 over water layer

- alkali metals
- alkaline earth metals
- transitional metals
- other metals
- nonmetals
- noble gases

58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium
90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium

Physical Science Lab Periodic Table

Katrin Zwirgmaier



Physical Science Lab Periodic Table:

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explained clearly and concisely Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory In today s classroom students not only need to understand the properties of the chemical elements but how these elements play such an integral role in industry the earth and the environment and in modern life No resource provides a better introduction than Robert Krebs s The History and Use of Our Earth s Chemical Elements In this thoroughly revised edition with extensive new and updated examples on the use of the chemical elements the elements are examined within their groups enabling students to make connections between elements of similar structure In addition the discovery and history of each element from those known from ancient times to those created in the modern laboratory is explained clearly and concisely In addition to the handy Guide to the Chemical Elements that comprises the bulk of the work The History and Use of Our Earth s Chemical Elements includes other useful features Introductory material on the basics of chemistry and the Periodic Table Appendices on the discoverers of the chemical elements A glossary of words commonly used in chemistry and chemical engineering A complete bibliography of useful resources including websites All of this information makes The History and Use of Our Earth s Chemical Elements the ideal one volume resource for understanding the importance of the chemical elements

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