

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts in the course. This packet highlights key concepts and their correct application in the course and should be used as a guide to focus on the material you should know.

B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

C. READING

1. Why do atoms bond together?

2. What is a bond?

3. What kind of electrons form covalent bonds?

4. What is an ionic bond?

5. What kind of electrons form ionic bonds?

6. Why don't noble gases bond together with other elements?

Why do noble gases not bond together with other elements?

1. Noble gases have a full outer shell of electrons.

2. Noble gases have a full outer shell of electrons.

Why do noble gases not bond together with other elements?

1. Noble gases have a full outer shell of electrons.

2. Noble gases have a full outer shell of electrons.

REACTIONS

What happens in a chemical reaction?

1. Reactants are combined to form products.

2. Reactants are combined to form products.

3. Reactants are combined to form products.

4. Reactants are combined to form products.

5. Reactants are combined to form products.

A chemical reaction is a process in which one or more substances are converted into one or more different substances. The substances that enter the reaction are called reactants, and the substances that are produced are called products.

1. What are the reactants in a chemical reaction?

Reactants are the substances that enter the reaction. They are usually written on the left side of the chemical equation, separated by plus signs.

2. What happens during the formation of a molecule?

During the formation of a molecule, atoms are joined together by chemical bonds. The bonds are formed by the sharing of electrons between the atoms.

3. What is a chemical reaction?

A chemical reaction is a process in which one or more substances are converted into one or more different substances.

4. What are the reactants in a chemical reaction?

Reactants are the substances that enter the reaction. They are usually written on the left side of the chemical equation, separated by plus signs.

5. What are the products in a chemical reaction?

Products are the substances that are produced by the reaction. They are usually written on the right side of the chemical equation, separated by plus signs.

6. Balance the following reactions:



B. VOCABULARY

1. Atom: A small particle consisting of a nucleus and electrons.

2. Atomic mass: The mass of an atom, measured in atomic mass units (amu).

3. Atomic weight: The average atomic mass of an element, measured in amu.

4. Chemical bond: A force that holds atoms together in a molecule or compound.

5. Chemical reaction: A process in which one or more substances are converted into one or more different substances.

6. Electronegativity: A measure of an atom's ability to attract electrons in a chemical bond.

7. Ion: A charged particle, either positive or negative, formed by the loss or gain of electrons.

8. Molecule: A group of two or more atoms held together by chemical bonds.

9. Reactant: A substance that enters a chemical reaction.

10. Product: A substance that is produced by a chemical reaction.

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RJ Alexander



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