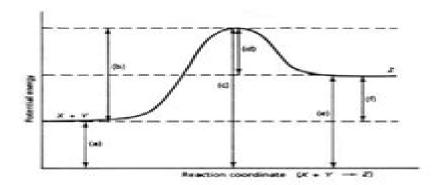
#### Potential Energy Diagram Worksheet Part I

- 1. Which of the letters a—f in the diagram represents the potential energy of the products?
- 2. Which letter indicates the potential energy of the activated complex?
- 3. Which letter indicates the potential energy of the reactants?
- 4. Which letter indicates the activation energy?
- 5. Which letter indicates the heat of reaction?
- 6. Is the reaction exothermic or endothermic?
- 7. Which letter indicates the activation energy of the reverse reaction?
- 8. Which letter indicates the heat of reaction of the reverse reaction?
- Is the reverse reaction exothermic or endothermic?

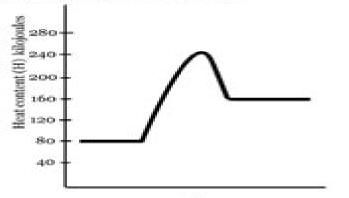




#### Potential Energy Diagram Worksheet Part II

- The heat content of the reactants of the forward reaction is about \_\_\_\_\_ kilojoules.
- The heat content of the products of the forward reaction is about kilojoules.
- The heat content of the activated complex of the forward reaction is about kJ
- The activation energy of the forward reaction is about \_\_\_\_\_ kilojoules.
- The heat of reaction (ΔH) of the forward reaction is about kilojoules.
- The forward reaction is \_\_\_\_\_\_
   (endothermic or exothermic).
- The heat content of the reactants of the reverse reaction is about \_\_\_\_\_kJ
- The heat content of the products of the reverse reaction is about \_\_\_\_\_ kilojoules.
- The heat content of the activated complex of the reverse reaction is about \_\_\_\_\_kJ.
- The activation energy of the reverse reaction is about \_\_\_\_\_ kilojoules.

- The heat of reaction (ΔH) of the reverse reaction is about \_\_\_\_\_ kilojoules.



Time

# **Potential Energy Diagram Worksheet**

**Michael Brown** 

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