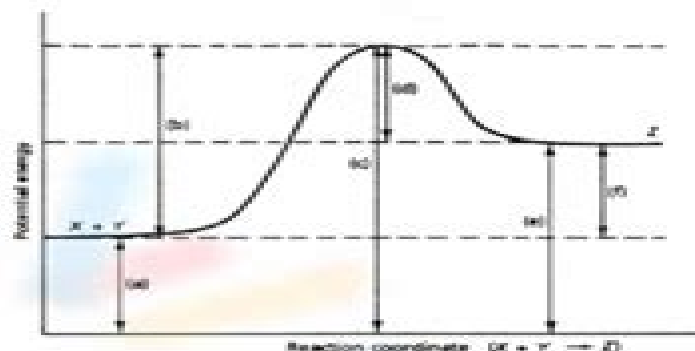
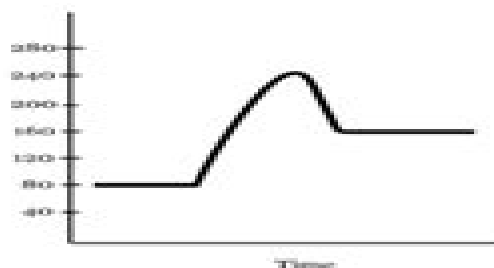


Potential Energy Diagram Worksheet **ANSWERS**



1. Which of the letters a-f in the diagram represents the potential energy of the products? **e**
2. Which letter indicates the potential energy of the activated complex? **c**
3. Which letter indicates the potential energy of the reactants? **a**
4. Which letter indicates the activation energy? **b**
5. Which letter indicates the heat of reaction? **f**
6. Is the reaction exothermic or endothermic? **endo**

7. Which letter indicates the activation energy of the reverse reaction? **d**
8. Which letter indicates the heat of reaction of the reverse reaction? **f**
9. Is the reverse reaction exothermic or endothermic? **exo**



(endothermic or exothermic).

1. The PE of the reactants of the forward reaction is about **80** kilojoules.
2. The PE of the products of the forward reaction is about **160** kilojoules.
3. The PE of the activated complex of the forward reaction is about **240** kilojoules.
4. The activation energy of the forward reaction is about **160** kilojoules.
5. The heat of reaction (ΔH) of the forward reaction is about **+80** kilojoules.
6. The forward reaction is **endothermic**
7. The PE of the reactants of the reverse reaction is about **160** kilojoules.
8. The PE of the products of the reverse reaction is about **80** kilojoules.
9. The PE of the activated complex of the reverse reaction is about **240** kilojoules.
10. The activation energy of the reverse reaction is about **80** kilojoules.
11. The heat of reaction (ΔH) of the reverse reaction is about **-80** kilojoules.
12. The reverse reaction is **exothermic** (endothermic or exothermic).

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