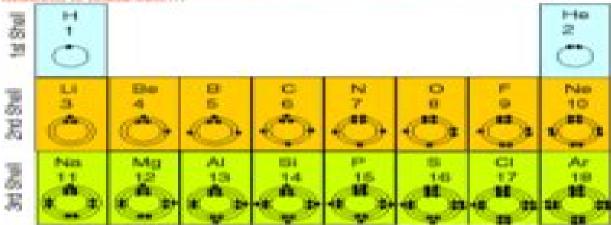
More Atomic Structure Practice: Use the periodic table to help you find the answers!

Draw atomic structures and electron dot diagrams for the following atomic H, He, Li, Be, B, C, O, F, Ne, Ne, Al, Si, P, Cl, Ar

Here are the atomic structures: NOTE! Only the electrons in their shells are shown below! The number of partiess equals the TOTAL number of electrons. The number of neutrons is found below.



Here are the electron det diagrams:

| 1 | 111 | | 133 | IV | W | VI | VIII | 0 |
|---|---------|--|--------|--------|--------|---------|------|------------|
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| | •Mg= | | - At - | - 10 - | -17 | 136 | 191 | 330 |
| | •Ca+ | | - 6 | 1000 | -Au | 175 | 100 | |
| | ette e | | - 10 | -5 | -60- | 17 | :00h | 200 |
| | - Out - | | 400 | -120- | | 2 750 - | 200 | |

How many neutrons are in the following atoms (show your work!): H. He, Li, Be, B. C., O. F. Ne, No. Al, St. P. Cl. Ar

Atomic mass (rounded) - Atomic Number = # neutrons

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K Payea

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