

Oxidation – Reduction Packet



Oxidation-reduction (redox) reactions are everywhere! They take place in your body, in plants, in the atmosphere, and in the oceans. They are the basis of many of the processes that sustain life on Earth.

Oxidation is the loss of electrons. Reduction is the gain of electrons. In a redox reaction, the oxidation state of one substance increases while the oxidation state of another substance decreases. The substance that is oxidized is the reducing agent, and the substance that is reduced is the oxidizing agent.

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Oxidation And Reduction Packet Key

Disha Experts

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the 'Disha Experts' text.

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