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temperature remains constant  $\Delta t = 0$  since internal energy depends on the temperature  $\Delta u = 0$  from first law of thermodynamics  
 $\Delta u = \Delta q + \Delta w$  since  $\Delta u = 0$   $\Delta q = -\Delta w$  also  $pV = nRT$  as  $T$  is constant  $pV = \text{constant}$