

Question	Answer	Mark	Guidance
2 (a)	<p>FIRST react all with</p> <p>Tollens' reagent AND silver mirror/ppt/solid (formed) with compound D</p> <p>OR with Fehling's/Benedict's solutions AND (brick-red/orange) solid/precipitate (formed) with compound D ✓</p> <p>NOTE: eliminates D</p> <div><chem>CC(O)CCC=O.[O]>>CC(O)CCC(=O)O</chem></div> <p>✓</p> <p>THEN react C and E with</p> <p>$\text{H}_2\text{SO}_4/\text{H}^+$ AND $\text{K}_2\text{Cr}_2\text{O}_7/\text{Cr}_2\text{O}_7^{2-}/\text{Na}_2\text{Cr}_2\text{O}_7$</p> <p>AND colour change OR green colour with compound C</p> <p>OR <u>no</u> change OR <u>no</u> reaction OR no green colour with compound E ✓</p> <div><chem>CC(=O)CCCC(=O)O.2[O]>>CC(=O)CCCC(=O)O.O</chem></div> <p>✓</p>	4	<p>ALLOW ammonia + silver nitrate for reagent</p> <p>ALLOW black solid/ppt</p> <p>ALLOW 'the aldehyde gives a silver mirror'</p> <p>ALLOW solid OR crystals OR ppt as alternatives for precipitate</p> <p>ALLOW correct structural OR displayed OR skeletal formulae</p> <p>OR combination of above as long as unambiguous</p> <p>DO NOT ALLOW molecular formulae for organic structures</p> <p>IGNORE all references to 2,4-dinitrophenylhydrazine/Brady's</p> <p>ACCEPT acidified dichromate</p> <p>ALLOW blue/green blue</p> <p>IGNORE equation for oxidation of D</p> <p>ALLOW equation for partial oxidation</p> <div><chem>CC(=O)CCCC(=O)O.O>>CC(=O)CCCC(=O)O.O</chem></div>

Ocr F324 June 2014 Question Paper

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