

Answer all the questions.

5. Methanoic acid, HCOOH , is a carboxylic acid that is present in some stings.

(a) What is the formula of the functional group that is responsible for the characteristic properties of carboxylic acids?

[2]

(b) Methanoic acid is usually referred to simply as formic acid.

Formic acid is made of carbon, hydrogen, oxygen and is insoluble in water.

Carboxylic acids react with carbonates in a similar way to other acids such as hydrochloric acid.



(c) Complete and balance this symbol equation for the reaction between calcium carbonate and methanoic acid.



[2]

(d) Calcium carbonate is insoluble so it does not react like this:

When calcium carbonate is heated, it reacts with methanoic acid, carbon methanoic acid.

The reaction with methanoic acid removes the calcium carbonate as insoluble. (Suggest a property of calcium methanoate that can explain why this happens.)

[2]

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