

EARTH AND LIFE SCIENCE
2ND QUARTER
G12 ABM- REVIEWER

LESSON 1:

Exogenic Processes

- It is a **blanket term** for those processes which operate on or close to the surface of the Earth.
- It includes **geological phenomena and processes** that originate externally to the Earth's surface.

Types of Exogenic Processes

Weathering

- It is the **breaking down of rocks** into fragments or into smaller size.
- Every rock had undergone weathering, whether it is **disintegration (physical weathering)** or **decomposition (chemical weathering)**.

Physical Weathering

- Refers to the **breakdown of rocks** into pieces without any change in its composition.

Types of Physical Weathering

Unloading

- It is the **uplifted intrusive igneous rocks**.
- It means rocks from **beneath the surface go above the surface**.
- **Weathering** are caused by **variety of factors** such as **tectonism and volcanism**.

Thermal Expansion & Contractions

- It refers to the **effect of temperature** to rocks.
- When exposed to **high temperature**, rocks **expand**; when exposed to **low temperature**, rocks **contract**.

Freeze-Thaw Weathering

- It is **most common** in the **North pole and South**.
- It occurs when a **fractured rock** absorbs **water** through its **crack** then **got frozen**.
- As mention in Thermal Expansion and contraction, the **colder the temperature**, the **molecule in it expand**.

Hydration

- It refers to the **amount of water** being **absorbed by the rock**.
- *When hydrated – too much or plenty of water is present – the rock softens.*
- *When it is dehydrated – too less or lack of water- the rock hardens.*
- The **repeated process** of this causes **weathering**

Salt Crystal Growth

- It usually **happens** in the **sea or the seaside**.
- It happens when **salt water** comes in **contact with rocks**.
- The **wind** can also take part in the **salt crystal growth**.

Chemical Weathering

- Refers to the **changes** in the **composition of rock**.

Types of Chemical Weathering

Oxidation

- O^2 in H^2O comes in **contact** with **certain elements** in **minerals of rocks**, and **chemical reaction** can occur.
- E.g. **Fe and Al** - $Fe^2 O^3$ and $Al^2 O^3$

Hydrolysis

- H^2O molecules alone or CO^2 in H^2O react with **chemical composition/components of rock minerals** to create **new components or compounds**.
- E.g. **silicate minerals (containing silica and O^2)** in **igneous rock** **silicate minerals – clay minerals**
 $= KaSi^2 O^2 + H^2O - Ha/Si$
- **Reactions** that produces **weak acid**.

Solution

- **Rock forming mineral** **dissolve** in **water**
- E.g. **halite mineral** containing **rock salt (NaCl)** **susceptible to solution** in H^2O

Carbonation

- Involves CO^2 and H^2O molecules **reacting** with, and thereby **decomposing** **rock material**.
- E.g. **carbonate rocks (CO^2) like limestone (composed of $CaCO^2$)** $(O^2 + H^2O - H^2 CO^2$ (carbonic acid) $CaCO$ (limestone) + $H^2 CO^2$ (carbonic acid) + $Ca HCO^2$) calcium bicarbonate

Mass Wasting

- also called **mass movement**
- **downslope transport** of **surface materials** in **direct response** to **gravity**.
- **heavier objects** have **greater downward materials** in **direct response** **gravity**.
- the **force of gravity** encourages **rock sediments** and **soil** to **move downhill or sloping surfaces**.

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