

Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes with pyridine ligands.

(a) The reaction shows two reactions of pyridine ligands with copper.

In the boxes, write the formulae and names of the complexes and ligands.

Complex	Ligand
$Cu(NH_3)_2$	$CuCl_2$
Blue	Yellow

(b)  $Cu^{2+}$  ions form a complex with 4 pyridine ligands and two water molecules. The 4 pyridine ligands are in a bidentate ligand.

The skeletal formula of the pyridine ligand is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term bidentate ligand?

A ligand with two donor atoms forming two coordinate bonds with central metal ion.

(d) The reaction is shown in Fig. 1.2 below.

The reaction is about the synthesis of copper complexes with pyridine ligands.



Complex A

Complete the 3D diagram of the other two pyridine ligands of A. You do not need to include any charges.

Indicate with ticks whether the stereocentres are chiral, achiral, optical or a combination of these types.

Structure														
Type	<table><tr><td>chiral</td><td></td></tr><tr><td>achiral</td><td><input checked="" type="checkbox"/></td></tr><tr><td>optical</td><td></td></tr></table>	chiral		achiral	<input checked="" type="checkbox"/>	optical		<table><tr><td>chiral</td><td><input checked="" type="checkbox"/></td></tr><tr><td>achiral</td><td></td></tr><tr><td>optical</td><td><input checked="" type="checkbox"/></td></tr></table>	chiral	<input checked="" type="checkbox"/>	achiral		optical	<input checked="" type="checkbox"/>
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(e) What is the empirical formula, including the charge, of the complex ion A?

$[Cu(NH_3)_4(H_2O)_2]^{2+}$

# Ocr Chemistry F32may 2013 Unofficial

**R Pring**



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