

SAMSUNG GALAXY S3 USER GUIDE MANUAL

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Samsung Galaxy S3 User Guide T Mobile

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ether solvent and the brown cloudy mixture was filtered by gravity filtration with fluted fast flow filter paper in a powder
funnel into a 250 ml round bottom flask rbf leaving light brown residue and beige filtrate diethyl ether was

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the trimyristin we performed a solid liquid extraction using the organic solvent methylene chloride CH_2Cl_2

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nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected

element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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