

Stoichiometry: Practice Assessment

Name: _____

5888 - Using a balanced equation, given the grams of reactant, calc. the grams of product



Balance the equation provided. Calculate the number of moles of sodium bromide that would be produced if 25.5 moles of gallium (III) bromide are used (with an excess of sodium sulfite)

$$25.5\text{mol GaBr}_3 \times 6\text{mol NaBr}/2\text{mol GaBr}_3 = 75.75\text{mol NaBr}$$

2) Tin (II) oxide reacts with nitrogen trifluoride to produce tin (II) fluoride and dinitrogen trioxide. Using the following reaction, calculate the grams produced of each product from 50.0 grams of nitrogen trifluoride.



$$50\text{g NF}_3 \times 1\text{mol NF}_3 / 71.001\text{g NF}_3 = 0.704\text{mol NF}_3 \times 2\text{mol SnF}_2 / 1\text{mol NF}_3 \times 156.69\text{g SnF}_2 / 2\text{mol SnF}_2 = 165.5\text{g SnF}_2$$

SnF₂

$$= 0.704\text{mol NF}_3 \times 1\text{mol N}_2\text{O}_3 / 2\text{mol NF}_3 \times 76.014\text{g N}_2\text{O}_3 / 1\text{mol N}_2\text{O}_3 = 26.75\text{g N}_2\text{O}_3$$

26.75g N₂O₃

3) A piece of copper wire is placed in silver nitrate. Silver and aqueous copper (II) nitrate is produced. The initial mass of the copper before the reaction was 2.74g and the mass remaining after the reaction was 2.32g. What is the mass of silver produced?



$$2.74\text{g} - 2.32\text{g} = 0.42\text{g}$$

$$0.42\text{g Cu} \times 1\text{mol Cu} / 63.55\text{g Cu} \times 2\text{mol Ag} / 1\text{mol Cu} \times 107.87\text{g Ag} / 1\text{mol Ag} = 1.43\text{g Ag}$$

5889 - After calculating theoretical yield in grams, calculate the percent yield. Given experimental data, determine the limiting reactant and the excess reactant

a) Through experimentation, you observe the reaction of iron (III) phosphate with sodium sulfate to make iron (III) sulfate and sodium phosphate. You began the experiment with 25g of iron (III) phosphate and 182g of sodium sulfate. How many grams of iron (III) sulfate should you have been able to produce? What is the percent yield if 18.5g of iron (III) sulfate are actually made?



$$25\text{g FePO}_4 \times 1\text{mol FePO}_4 / 150.815\text{g FePO}_4 = 0.166\text{mol FePO}_4 / 2\text{mol per rxn} = 0.083\text{rxns (LR)}$$

$$182\text{g Na}_2\text{SO}_4 \times 1\text{mol Na}_2\text{SO}_4 / 142.04\text{g Na}_2\text{SO}_4 = 1.28\text{mol Na}_2\text{SO}_4 / 3\text{mol per rxn} = 0.427\text{rxns}$$

$$0.083\text{mol FePO}_4 \times 1\text{mol Fe}_2(\text{SO}_4)_3 / 2\text{mol FePO}_4 \times 399.873\text{g Fe}_2(\text{SO}_4)_3 / 1\text{mol Fe}_2(\text{SO}_4)_3 = 33.19\text{g}$$

$$(18.5\text{g} / 33.19\text{g}) \times 100 = 56\%$$

Review Stoichiometry Section 2 Answer Key

Arshad Iqbal



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, DAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests Kaplan Test Prep, 2016-10-04 Kaplan's DAT 2017 2018 Strategies Practice Review features the realistic practice test taking strategies and expert guidance you need to score higher on the Dental Admissions Test Our comprehensive subject review and test blueprint will help you face Test Day with confidence The Best Review Two full length online practice tests More than 600 practice questions for every subject with detailed answers and explanations 12 page tear out full color study sheets for quick review on the go A guide to the current DAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the DAT Biology General Chemistry Organic Chemistry Perceptual Ability Reading Comprehension Quantitative Reasoning Kaplan's proven strategies for Test Day success Expert Guidance Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test We invented test prep Kaplan www kaptest com has been helping students for almost 80 years Our proven strategies have helped legions of students achieve their dreams OAT 2017-2018 Strategies, Practice & Review with 2 Practice Tests Kaplan Test Prep, 2016-10-04 Kaplan's OAT 2017 2018 Strategies Practice Review provides the content review test taking strategies and realistic practice you need to get the OAT results you want Updated for the latest test changes OAT 2017 2018 is your guide to facing Test Day with confidence The Best Review Two full length online practice tests More than 600 practice questions for every subject with detailed answers and explanations 16 page tear out full color study sheets for quick review on the go A guide to the current OAT Blueprint so you know exactly what to expect on Test Day Comprehensive review of all of the content covered on the OAT Biology General Chemistry Organic Chemistry Reading Comprehension Physics Quantitative Reasoning Kaplan's proven strategies for Test Day success Expert Guidance Kaplan's expert psychometricians ensure our practice questions and study materials are true to the test We invented test prep Kaplan www kaptest com has been helping students for almost 80 years Our proven strategies have helped legions of students achieve their dreams *Holt Chemistry* Ralph Thomas Myers, 2004 **E3 Chemistry Review Book - 2018 Home Edition (Answer Key Included)** Effiong Eyo, 2017-10-20 With Answer Key to All Questions Chemistry students and homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Review Book 2018 With E3 Chemistry Review Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with solutions to study and follow Several practice multiple choice and short answer questions at the end of each lesson to test understanding of the materials 12 topics of Regents question sets and 3 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 197836229 The Home Edition contains an answer key section Teachers who want to

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