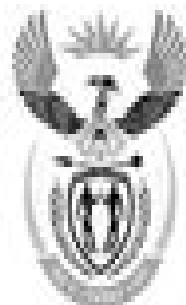


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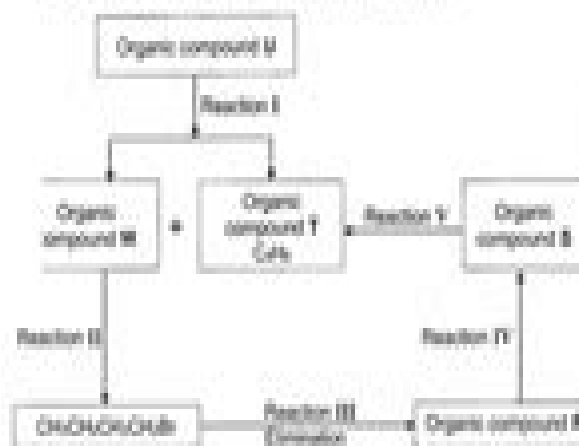
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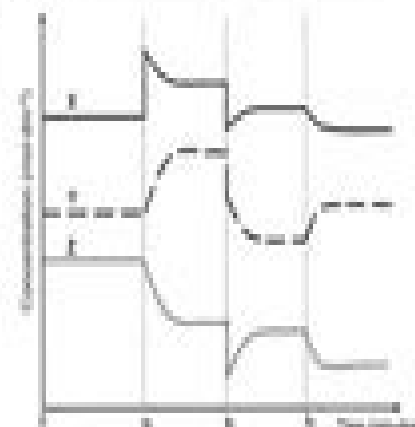
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Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

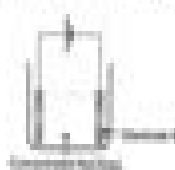
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is an oxidation reaction.



4.8 Write the half-cell reaction for the anode in the above cell.

4.9 Write the overall cell reaction for the above cell.

5.1 Write down the name of the reaction in the above diagram.

5.2 Write down the name of the reaction in the above diagram.



5.3 Write down the name of the reaction in the above diagram.

5.4 Write down the name of the reaction in the above diagram.

5.5 Write down the name of the reaction in the above diagram.

5.6 Write down the name of the reaction in the above diagram.

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