

Memorandum Grade 11 p/sciences

Physical Sciences Gr11/P2

2
NSC

September

QUESTION 1

1.1 A ✓✓

(2)

1.2 B ✓✓

(2)

1.3 C & ✓✓

(2)

1.4 B & ✓✓

(2)

1.5 C & ✓✓

(2)

1.6 D ✓✓

(2) ~~(10)~~ (12)

QUESTION 2

2.1 The mass of one mole of a substance measured in $\text{g} \cdot \text{mol}^{-1}$ ✓✓ (2 or 0)

(2)

2.2

Marking criteria		
(a) Any formula $m = nM$ or $N = nN_A$ ✓		
(b) Substitution of $3,114 \times 10^{24}$ and $6,02 \times 10^{23}$ into $N = nN_A$ ✓		
(c) Substitution of $1 \text{ g} \cdot \text{mol}^{-1}$ into $m = nM$ ✓		
(d) Mole ratio 3:6:1 ✓		
(e) $X = 3$ ✓, $Y = 6$ ✓, $Z = 1$ ✓		
C	H	O
$n = \frac{N}{N_A}$ ✓ $n = \frac{3,114 \times 10^{24}}{6,02 \times 10^{23}}$ ✓ $n = 5,173 \text{ mol}$	$n = \frac{m}{M}$ $n = \frac{10,345}{1}$ ✓ $n = 10,345 \text{ mol}$	$n = 1,724 \text{ mol}$
$\frac{5,173}{1,724} = 3$	$\frac{10,345}{1,724} = 6$	$\frac{1,724}{1,724} = 1$
3:6:1 ✓		
$X = 3$ ✓	$Y = 6$ ✓	$O = 1$ ✓

Accept values indicated in the empirical formula: $\text{C}_3\text{H}_6\text{O}$ ✓✓✓

(7)

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R Barnett



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Table of Contents Physical Science Memoradam Grade 11 September 2013

1. Understanding the eBook Physical Science Memoradam Grade 11 September 2013
 - The Rise of Digital Reading Physical Science Memoradam Grade 11 September 2013
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Memoradam Grade 11 September 2013
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Memoradam Grade 11 September 2013
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Memoradam Grade 11 September 2013
 - Personalized Recommendations
 - Physical Science Memoradam Grade 11 September 2013 User Reviews and Ratings
 - Physical Science Memoradam Grade 11 September 2013 and Bestseller Lists
5. Accessing Physical Science Memoradam Grade 11 September 2013 Free and Paid eBooks
 - Physical Science Memoradam Grade 11 September 2013 Public Domain eBooks
 - Physical Science Memoradam Grade 11 September 2013 eBook Subscription Services
 - Physical Science Memoradam Grade 11 September 2013 Budget-Friendly Options

6. Navigating Physical Science Memoradam Grade 11 September 2013 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Memoradam Grade 11 September 2013 Compatibility with Devices
 - Physical Science Memoradam Grade 11 September 2013 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Memoradam Grade 11 September 2013
 - Highlighting and Note-Taking Physical Science Memoradam Grade 11 September 2013
 - Interactive Elements Physical Science Memoradam Grade 11 September 2013
8. Staying Engaged with Physical Science Memoradam Grade 11 September 2013
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Memoradam Grade 11 September 2013
9. Balancing eBooks and Physical Books Physical Science Memoradam Grade 11 September 2013
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Memoradam Grade 11 September 2013
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Memoradam Grade 11 September 2013
 - Setting Reading Goals Physical Science Memoradam Grade 11 September 2013
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Memoradam Grade 11 September 2013
 - Fact-Checking eBook Content of Physical Science Memoradam Grade 11 September 2013
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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