

Name _____

Physical Science Chemical Bonding Study Guide

A. Covalent & Ionic Compounds/Bonds

1. Which ion always forms with a charge of 2-? _____ (oxygen)
which cation is always a 2+? _____ (magnesium)
2. How many valence electrons does _____ (oxygen) have? _____ (6)
How many valence electrons does _____ (magnesium) have? _____ (2)
3. How many _____ (electrons) does _____ (oxygen) have? _____ (6)
How many _____ (electrons) does _____ (magnesium) have? _____ (2)
4. How many _____ (electrons) does _____ (oxygen) have? _____ (6)
How many _____ (electrons) does _____ (magnesium) have? _____ (2)
5. How many _____ (electrons) does _____ (oxygen) have? _____ (6)
How many _____ (electrons) does _____ (magnesium) have? _____ (2)
6. How many _____ (electrons) does _____ (oxygen) have? _____ (6)
How many _____ (electrons) does _____ (magnesium) have? _____ (2)
7. How many _____ (electrons) does _____ (oxygen) have? _____ (6)
How many _____ (electrons) does _____ (magnesium) have? _____ (2)

B. Formulas for Ionic Compounds - Fill in the blank below

Write the correct chemical formula for the compound.	Aluminum	Fluorine	Phosphorus	Sulfur
_____	_____	_____	_____	_____

C. Write the chemical formula for each of the following using the periodic table

Write your answer below the element symbols, and use arrows to show the transfer of electrons. Use the periodic table to find the oxidation numbers. They will usually be the same as the element's group number, except for the transition metals. Use the same method to find the oxidation numbers for the other elements.

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