

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
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Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

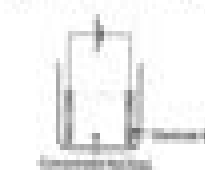
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction type for Reaction I.



4.8 Calculate the standard cell potential for the cell shown above.

Use the standard electrode potentials of the products.

4.9 Calculate the standard cell potential for the cell shown above.



4.10 The simplified diagram below represents an electrochemical cell. The electrodes are made of metal.

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Albert A Gayle



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