

STUDY GUIDE 11: Oxidation and Reduction (Chapter 12)

Remember: Oxidation = Loss of Electrons

Oxidation: loss of electrons; decrease in O.N. number (including 0, 0, 0 bonds)

Reduction: gain of electrons; increase in O.N. number; decrease in O.N. number



Be careful: some compounds contain both oxidized and reduced carbon atoms. For example:



REDOX REACTIONS Redox reactions involve the

Transfer of

Electrons. For example, the reaction between zinc and copper(II) ions is a redox reaction.



Therefore, redox reactions are reactions in which electrons are transferred from one substance to another.



Oxidation And Reduction Study Guide

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Oxidation And Reduction Study Guide:

Unveiling the Power of Verbal Art: An Emotional Sojourn through **Oxidation And Reduction Study Guide**

In some sort of inundated with displays and the cacophony of instant conversation, the profound power and emotional resonance of verbal artistry usually diminish into obscurity, eclipsed by the regular onslaught of noise and distractions. However, set within the lyrical pages of **Oxidation And Reduction Study Guide**, a interesting perform of fictional brilliance that pulses with raw emotions, lies an wonderful trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this enchanting opus guides visitors on a psychological odyssey, softly exposing the latent possible and profound influence embedded within the complicated internet of language. Within the heart-wrenching expanse of the evocative evaluation, we can embark upon an introspective exploration of the book is main themes, dissect its interesting publishing style, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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