

**Max. Marks:70**

**Time: 3 hours**

**General Instructions:**

**Read the following instructions carefully.**

- (a) There are 33 questions in this question paper with internal choice.
  - (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
  - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
  - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
  - (e) SECTION D consists of 2 case - based questions carrying 4 marks each.
  - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
  - (g) All questions are compulsory.
  - (h) Use of log tables and calculators is not allowed.
- 

### SECTION A

*The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.*

1. Which of the following solutions will have the highest conductivity at 298 K?
 

(a) 0.01 M HCl solution	(b) 0.1 M HCl solution
(c) 0.01 M CH <sub>3</sub> COOH solution	(d) 0.1 M CH <sub>3</sub> COOH solution



Identify A and B:

- |  |   |
|--|---|
| (a) A = 1-phenylethanal , B = acetophenone | (b) A = Benzophenone B = formaldehyde   |
| (c) A = Benzaldehyde , B = Acetophenone    | (d) A = Benzophenone , B = Acetophenone |
3. The vitamins which can be stored in our body are:
 

(a) Vitamin A, B, D and E	(d) Vitamin A, C, D and K
(c) Vitamin A, B, C and D	(d) Vitamin A, D, E and K
  4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give CH<sub>3</sub>CH= C(CH<sub>3</sub>)COONa and yellow precipitate of CHI<sub>3</sub>?
 

(a) 3-Methylpent-3-en-2-one	(b) 3-Methylbut-2-en- one
(c) 2, 3-Dimethylethanone	(d) 3-Methylpent-4-one
  5. Which of the following is not correct?
 

(a) In haloarenes, the electron pairs on halogen atom are in conjugation with π-electrons of the ring.	(b) The carbon-magnesium bond is covalent and non-polar in nature .
(c) During S <sub>N</sub> <sup>1</sup> reaction, the carbocation formed in the slow step being sp <sup>2</sup> hybridised is planar.	(d) Out of CH <sub>2</sub> = CH-Cl and C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Cl, C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Cl is more reactive towards S <sub>N</sub> <sup>1</sup> reaction

# Sample Question Paper Applied Chemistry

**N Colangelo**



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## Embracing the Tune of Term: An Mental Symphony within **Sample Question Paper Applied Chemistry**

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