

Results

and

Discussion



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RESEARCH ARTICLE

Application of Group I Metal Addition to the Separation of Steroids by Traveling Wave Ion Mobility Spectrometry

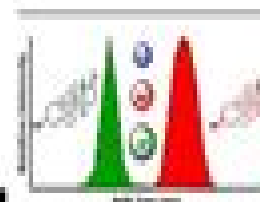


Figure 1. Traveling wave ion mobility spectrometry (TWIM) spectra showing the separation of steroids. The x-axis is labeled 'Ion Mobility' and the y-axis is labeled 'Relative Abundance'. Three distinct peaks are shown: a green peak labeled 'G', a red peak labeled 'R', and a blue peak labeled 'B'. The peaks are separated by their ion mobility, with G being the most mobile and B being the least mobile.

Traveling wave ion mobility spectrometry (TWIM) was applied to the separation of steroids. The results showed that the addition of group I metal ions (Li⁺, Na⁺, K⁺, Rb⁺, Cs⁺) to the steroid solutions significantly improved the separation efficiency. The separation efficiency was evaluated by the resolution (R_s) and the peak area ratio (P_r). The results showed that the addition of group I metal ions significantly improved the separation efficiency, with R_s values ranging from 1.5 to 2.5 and P_r values ranging from 0.5 to 1.0.

Keywords: Traveling wave ion mobility spectrometry, Steroids, Group I metal ions, Separation efficiency

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Introduction

Traveling wave ion mobility spectrometry (TWIM) is a powerful analytical tool for the separation and identification of complex mixtures. It has been widely used in the field of mass spectrometry for the separation of various compounds, including steroids. The addition of group I metal ions to the steroid solutions has been shown to significantly improve the separation efficiency.

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