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8 3 concentrations of solutions problems chemistry libretexts - Dec 13 2022

web the hardness of water hardness count is usually expressed in parts per million by mass of CaCO_3 which is equivalent to milligrams of CaCO_3 per liter of water what is the molar concentration of Ca^{2+} ions in a water sample with a hardness count of 175 mg CaCO_3 l

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it's written like this $\frac{\text{begin array l mass of a component}}{\frac{\text{frac mass of component in solution}}{\text{total mass of solution}}} \times 100$ end array

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams