

1. The basic premise of the collision theory is that particles must first _____ if a reaction is to occur. Additionally, that _____ must be effective. To be effective, the particles must _____ with the _____ and the proper amount of _____.
2. _____ is the study of reaction rates which is the rate at which the _____ disappear and the products _____.
3. There are five factors affecting the reaction rate:
 - a. _____ of the _____
 - b. _____ of the _____
 - c. _____
 - d. _____
 - e. _____
4. Use the collision theory to explain:
 - a. A mixture of hydrogen gas and chlorine gas reacts faster when the volume they occupy is decreased: _____

 - b. Iron filings rust faster than iron wire: _____

 - c. An increase in temperature increases the reaction rate: _____

5. Why do you think that industrial chemists are concerned with reaction rates?

Review Sheet Unit 12 Chemistry Answers

Annelies Wilder-Smith



Review Sheet Unit 12 Chemistry Answers:

Decoding **Review Sheet Unit 12 Chemistry Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Review Sheet Unit 12 Chemistry Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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