

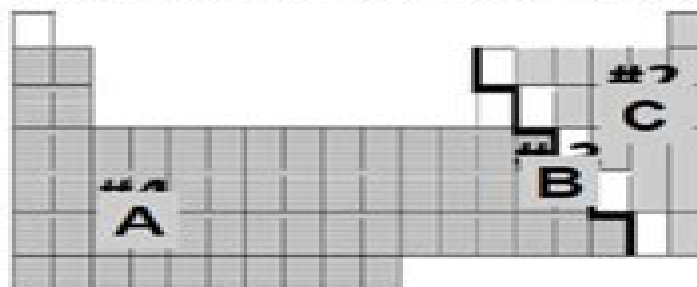
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Review: Unit 4 Periodic Table

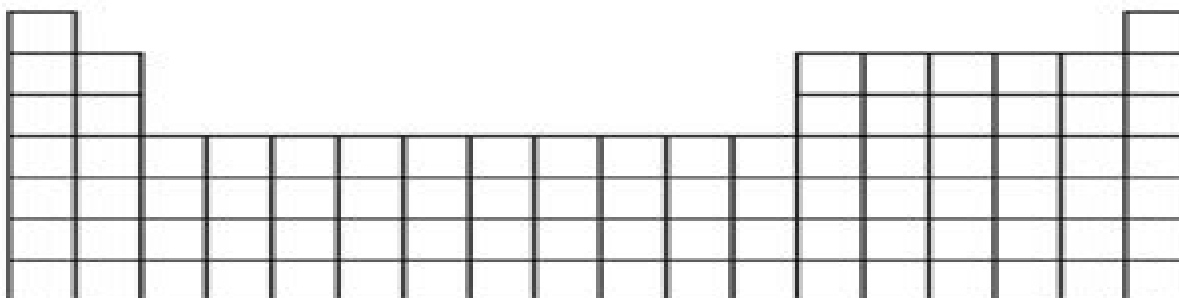
Part 1: History of the Periodic Table/ Families of the PT

1. Dimitri Mendeleev put elements in order based on _____
2. Henry Mosely put elements in order based on _____
3. Label the 3 sections of this periodic table as (Metalloid, Metals, Non-Metals)



Part 2: Periodic Trends

4. Label the periodic table with the following periodic table trends: Atomic Radius, Ionization Energy, Electronegativity (Use different colors for each trend)



5. Where on the Periodic Table is Electronegativity zero?

Part 3: Average Atomic Mass

6. What is the unit for average atomic mass? What does this unit mean, how is it measured?
7. 24.1% of all the isotopes of an element have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu. What is the average mass of this element?

Review The Periodic Table Section

Chao Zhang



Review The Periodic Table Section :

Reviewing **Review The Periodic Table Section** : Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Review The Periodic Table Section** ," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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