

GRADE 10 PHYSICAL SCIENCES

NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

SECTION A
QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A. distilled water
- B. diamond
- C. sodium chloride
- D. ketchup

1.2 Element with the electron configuration: $1s^2, 2s^2, 2p^4$

- A. oxygen
- B. carbon
- C. helium
- D. nitrogen

1.3 The chemical formula of sodium sulfide

- A. NaS
- B. Na₂S
- C. NaS₂
- D. Na₂S₂

1.4 Strongest intermolecular force found in water?

- A. Ion-dipole
- B. Dipole-dipole
- C. Van der Waals
- D. Hydrogen bond

1.5 The particles of a gas ...

- A. are tightly packed.
- B. vibrate against each other.
- C. move freely and rarely touch.
- D. move over one another.

1.6 Element with the highest electronegativity?

- A. Hydrogen
- B. Helium
- C. Oxygen
- D. Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A. mass number of the atom will increase
- B. atomic number of the atom will decrease
- C. number of electrons will decrease
- D. number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a:



- A. noble gas
- B. halogen
- C. alkali earth-metal
- D. alkali metal

Physical Science Grade 10 November 2013 Paper 2

R. Burke Johnson, Larry B. Christensen

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