Blackline Master 5.8

# Ionic Compounds: Names and Formulas Worksheet

(a) magn	esium oxide	Mg0	(k) copper(i) bron	
(b) sodium	m fluoride	NaP	(I) tin(II) iodide	Sn Brz
(c) alumir	num nitride	AIN	(m) iron(III) chlori	deFECI3
(d) potass	sium sulfide	K25	(n) calcium phosp	hide _ Caz Pz
(e) lithium	niodide	LiT	(o) lead(II) cxide	Pbo
(f) calcium bromide		Cabra	(p) lead(IV) fluorid	e PbF4
g) beryllis	um oxide	BeO	(q) tin(IV) bromide	Sn Bry
(h) nickel	chloride	Nio	(r) copper(II) sulfid	1 -
i) magnes	sium nitride	MgzNz	(s) iron(II) oxide	FeO
(i) eluminum sulfide			o sere and an experience of	
Write the r	names for the f	AlsS3	(t) calcium nitride	Ca3 N2
Write the r	names for the f	ollowing compounds.	(k) PbS	ead (II) su(ficle
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub>	lithium aluminu	ollowing compounds. oxide un chloride	(k) PbS	ead (II) su(ficle in (IV) oxide
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub>	lithium aluminu Magnes	ollowing compounds. oxide un chloride ium sulfide	(k) PbS(c) SnO <sub>2</sub> +	ead (II) su(ficle in (IV) oxide dium sulficle
Write the r a) Li <sub>2</sub> O b) AICI <sub>3</sub> c) MgS	lithium aluminu magnes	ollowing compounds. oxide un chloride ium sulfide m oxide	(k) PbS(c) (l) SnO <sub>2</sub> + (m) Na <sub>2</sub> S	ead (II) su(ficle in (II) oxide dium sulficle agnosium phosphic
Write the r a) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO	names for the f lithium aluminu magnes calcius	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide	(k) PbS	ead (II) su(ficle in (IV) oxide dium sulficle
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO	names for the f lithium aluminu magnes calcius	ollowing compounds. oxide un chloride ium sulfide m oxide	(k) PbS(c) (l) SnO <sub>2</sub>	ead (II) su(ficle in (II) oxide dium sulficle agnosium phosphic
	mames for the f lithium aluminu magnes calcius potassi beryllic	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide	(k) PbS	ead (II) su(ficle in (ID) oxide fium sulficle agresium phosphic ckel (II) oxide
Write the r (a) Li <sub>2</sub> O (b) AICI <sub>3</sub> (c) MgS (d) CaO (e) KBr	magras calcius potassi beryllic	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide um fluoride	(k) PbS	ead (II) su(ficle in (IV) oxide dium sulficle agnosium phosphic ckel (II) oxide opper (I) iodide
Write the r a) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO e) KBr	magnes calcius pofassi benyllic sodius	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide um fluoride n ni fride	(k) PbS	ead (II) sulficle in (IV) oxide dium sulficle agnosium phosphic ckel (II) oxide opper (I) iodicle ad (IB) chloricle

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