2.1

PROPERTIES OF MATTER

Section Review

Objectives

- · Identify physical properties and physical changes
- Distinguish intensive properties from extensive properties
- Differentiate among three states of matter

Vocabulary

- IIIIANSS
- volume
- extensive property
- · intensive property

- substance.
- · physical property
- scolid.
- Biquid

- 13 games
- vapor
- · physical change

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Properties used to describe matter can be classified as

1 or 2 . The 3 of an object is a measure of

the amount of matter the object contains. The ____4___ of an object is a measure of the space occupied by the object. An extensive

property is one that depends on the 5 of matter. An intensive

property is one that depends on the ____6__ of matter.

A 7 is matter that has uniform and definite composition.

A solid has a definite _____8 and ____9 ___. A liquid has a definite

volume, but takes the _____ of its container. A ____ takes

both the shape and volume of its container.

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

11. Matter has mass and occupies space.

12. A liquid has a definite shape.

13. Heating a solid to 200°C will cause it to change to a liquid.

______ 14. Gases are easier to compress than liquids.

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