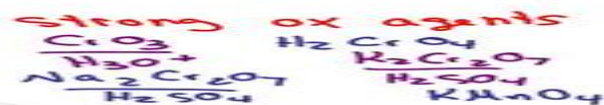


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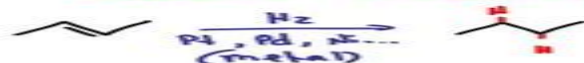
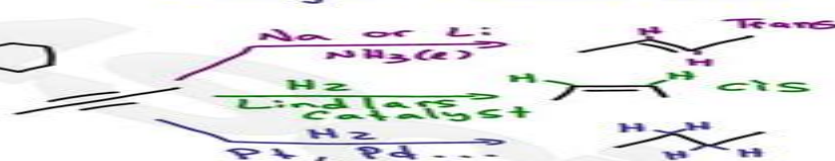
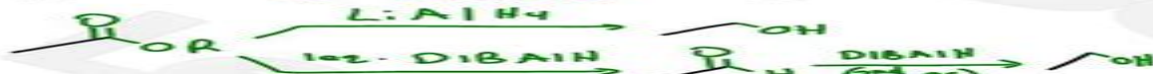
CC=CC
 $\xrightarrow[\text{KOH, } \text{O}_4]{\text{OsO}_4}$
CC(O)CC(O)





only 1° & 2°
get oxidized
3° = N/A

Alkene Reduction

c1ccccc1 $\xrightarrow[\text{ROH, NH}_3]{\text{Li or Na}}$ c1ccccc1CC(=O)O $\xrightarrow{LiAlH_4}$ CCO

$\text{Li}^+ + \text{H}-\overset{\text{H}}{\underset{\text{H}}{\text{Al}}}-\text{H}$
 $\text{NaOH} + \text{H}-\overset{\text{H}}{\underset{\text{H}}{\text{Al}}}-\text{H} = \text{weak}$



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Oxidation Reduction Reactions Study Guide

Chao Zhang



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