

Ch. 9 Plate Tectonics: Study Guide

Vocabulary

9.4 Earth's Layered Structure: crust, p. 233; mantle, p. 234; lithosphere, p. 234; asthenosphere, p. 235; outer core, p. 235; inner core, p. 235;

9.1 Plate Tectonics: continental drift, p. 248; Pangaea, p. 248; plate tectonics, p. 254; divergent boundary, p. 255; convergent boundary, p. 255; transform fault boundary, p. 255; ridge, p. 258; seafloor spreading, p. 259; subduction zone, p. 261; trench, p. 261; paleomagnetism, p. 265; hot spot, p. 268

Reviewing Content

Choose the letter that best answers the question or completes the statement.

- What is the weaker, hotter zone beneath the lithosphere that allows for motion of Earth's rigid outer shell?
 a. Crust
b. Asthenosphere
 c. outer core
 d. inner core
- Most of Earth's earthquakes, volcanoes, and mountain building occur
 a. in the center of continents.
 b. in the Himalayas.
c. at plate boundaries.
 d. at volcanic island arcs.
- Alfred Wegener is best known for what hypothesis?
 a. plate tectonics
 b. seafloor spreading
c. continental drift
 d. subduction
- Complex mountain systems such as the Himalayas are the result of
 a. oceanic-oceanic convergence.
 b. hot spots.
c. continental volcanic arcs.
d. continental-continental convergence.
- What is the type of plate boundary where two plates move together, causing one of the slabs of lithosphere to descend into the mantle beneath an overriding plate?
a. oceanic-continental convergent
 b. Divergent
 c. transform fault
 d. continental-continental convergent
- One of the main objections to Wegener's hypothesis of continental drift was that he was unable to provide an acceptable
 a. rate of continental drift.
 b. date of continental drift.
c. mechanism of continental drift.
 d. direction of continental drift.
- Which one of the following was NOT used as support of Wegener's continental drift hypothesis?
a. fossil evidence
b. Paleomagnetism
 c. the fit of South America and Africa
 d. ancient climates
- At what type of plate boundary do plates move apart, resulting in the upwelling of material from the mantle to create new seafloor?
a. Divergent
 b. Convergent
 c. transform fault
 d. Subduction

Use the diagram below to answer Questions

- What feature is labeled B?
b. a trench
 a. continental volcanic arc
 c. continental lithosphere
 d. an ocean ridge
- The process occurring at the location labeled D is
 a. oceanic lithosphere being created.
 b. continental lithosphere being created.
c. a continental-continental collision occurring.
d. oceanic lithosphere being subducted.

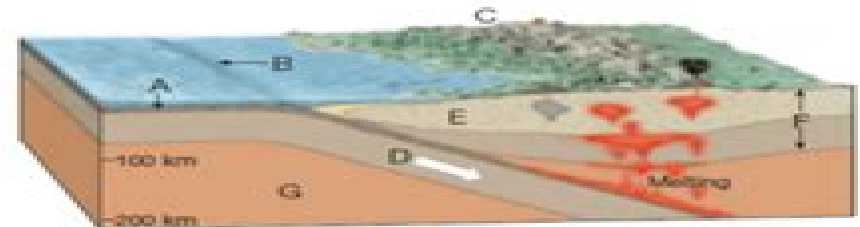


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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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