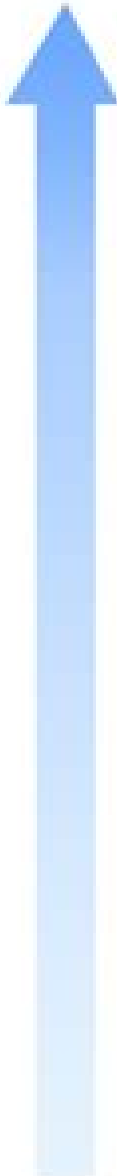


Reactivity Series

Metal (Ion)	Reactivity	Most reactive
Potassium (K ⁺)	Reacts with water	
Sodium (Na ⁺)		
Lithium (Li ⁺)		
Barium (Ba ²⁺)		
Strontium (Sr ²⁺)		
Calcium (Ca ²⁺)		
Magnesium (Mg ²⁺)	Reacts with acid	
Aluminum (Al ³⁺)		
Manganese (Mn ⁺)		
Zinc (Zn ²⁺)		
Chromium (Cr ²⁺)		
Iron (Fe ²⁺)		
Cadmium (Cd ²⁺)		
Cobalt (Co ²⁺)		
Nickle (Ni ²⁺)		
Tin (Sn ²⁺)		
Lead (Pb ²⁺)		
Hydrogen (H ⁺)	Included for comparison	
Antimony (Sb ²⁺)	Highly unreactive	
Bismuth (Bi ²⁺)		
Copper (Cu ²⁺)		
Mercury (Hg ²⁺)		
Silver (Ag ⁺)		
Gold (Au ²⁺)		
Platinum (Pt ⁺)	Least reactive	

Reactivity Series Pogil

Huangqi Zhang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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