

Name:		
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I. Single Replacement Reactions

(Remember, some reactions will NOT proceed. If there is no reaction, write "NR" to the right of the arrow.)

- Write the reaction for each of the following reactions:
 - a) Sodium sulfate reacts with magnesium to produce magnesium sulfate and sodium
 - b) Copper(I) carbonate reacts with aluminum to form copper metal and aluminum carbonate.
- 2. Complete the following reactions:
 - a) Ca(NO₃)₂ + AI →
 - b) Zn + K₃PO₄ →

II. Double Replacement Reactions

- 3. Write the reaction for each of the following reactions:
 - a) Iron(II) phosphate reacts with sodium chloride to produce iron(II) chloride and sodium phosphate.
 - b) Calcium carbonate reacts with lithium hydroxide to produce calcium hydroxide and lithium carbonate.
- 4. Complete the following reactions:
 - a) Mg(NO₃)₂ + Al(C₂H₃O₂)₅ →
 - b) Ba(OH)₂ + Na₃PO₄ →

III. Combustion Reactions

You can try the following formula to balance: $(CxHy + (x + y/4)O_2 \rightarrow xCO_2 + yH_2O)$

- Write the combustion reactions for the following compounds.
 - a) C₂H₄
 - b) C₈H₁₆

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Table of Contents Reactions Practice Skill Practice 28

- 1. Understanding the eBook Reactions Practice Skill Practice 28
 - The Rise of Digital Reading Reactions Practice Skill Practice 28
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Reactions Practice Skill Practice 28
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Reactions Practice Skill Practice 28
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Reactions Practice Skill Practice 28
 - Personalized Recommendations
 - Reactions Practice Skill Practice 28 User Reviews and Ratings
 - Reactions Practice Skill Practice 28 and Bestseller Lists
- 5. Accessing Reactions Practice Skill Practice 28 Free and Paid eBooks
 - Reactions Practice Skill Practice 28 Public Domain eBooks
 - Reactions Practice Skill Practice 28 eBook Subscription Services
 - Reactions Practice Skill Practice 28 Budget-Friendly Options
- 6. Navigating Reactions Practice Skill Practice 28 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Reactions Practice Skill Practice 28 Compatibility with Devices
 - Reactions Practice Skill Practice 28 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Reactions Practice Skill Practice 28
 - Highlighting and Note-Taking Reactions Practice Skill Practice 28
 - Interactive Elements Reactions Practice Skill Practice 28
- 8. Staying Engaged with Reactions Practice Skill Practice 28
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Reactions Practice Skill Practice 28
- 9. Balancing eBooks and Physical Books Reactions Practice Skill Practice 28
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Reactions Practice Skill Practice 28
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - $\circ \ \ Managing \ Screen \ Time$
- 11. Cultivating a Reading Routine Reactions Practice Skill Practice 28
 - Setting Reading Goals Reactions Practice Skill Practice 28
 - Carving Out Dedicated Reading Time

- 12. Sourcing Reliable Information of Reactions Practice Skill Practice 28
 - Fact-Checking eBook Content of Reactions Practice Skill Practice 28
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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