

Equilibrium and Reaction Rates

(answers at the end)

1. What is Le Chatelier's law (Le Chatelier's principle)?

It describes a system's behaviour in response.

It states that if a system is at equilibrium and a change is made to it, the system will adjust itself to counteract the change.

For example, if a system is at equilibrium and a change is made to it, the system will adjust itself to counteract the change. If a system is at equilibrium and a change is made to it, the system will adjust itself to counteract the change.

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For example, if a system is at equilibrium and a change is made to it, the system will adjust itself to counteract the change.

- (a) the system is at equilibrium
- (b) the system is at equilibrium
- (c) the system is at equilibrium
- (d) it is at equilibrium in the absence of any change
- (e) the system is at equilibrium

4. What is Le Chatelier's law (Le Chatelier's principle)?

It describes a system's behaviour in response.

Reactions Rates And Equilibrium Section Review Answers

**Louis Theodore, R. Ryan Dupont, Kumar
Ganesan**



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Systems and Materials Nicolas Giuseppone, Andreas Walther, 2021-07-19 Out of Equilibrium Supra molecular Systems and Materials A must have resource that covers everything from out of equilibrium chemical systems to active materials Out of

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