Name	answer key	Date
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TYPES OF CHEMICAL REACTIONS

Complete each of the following equations as needed to make it the type of reaction indicated.

Be sure to write each formula correctly.

Double replacement:

 $Na_2CrO_4 + PbCl_2$

2 Nacl + PbCrO₄

2) Single replacement:

 $Cl_2 + 2 Nabr$

2 NaCl + Br2

3) Decomposition:

 $Mg(ClO_3)_2$

2 H₂O

4) Synthesis:

 $2 H_2 + O_2$

2 Fe(OH)3 + 3 CaCl2

5) Double replacement:

3 Ca (OH)2 + 2 FeCl3

2 H₂O

6) Single replacement:

 $Fe + Cu(NO_3)_2$

 $Fe(NO_3)_2 + Cu$

7) Decomposition:

 $2 \text{ Hg}_2 O$

 $4Hg + O_2$

8) Synthesis:

 $S + O_2$

 50_2

9) Double replacement:

AgNO₃ + KI

Agl + KNO₃

10) Single replacement:

Cu + 2 AgNO₃

 $2 \text{ Ag} + \text{Cu}(\text{NO}_3)_2$

Reactions Classification Answer Key

Oswaal Editorial Board

Reactions Classification Answer Key:

The book delves into Reactions Classification Answer Key. Reactions Classification Answer Key is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Reactions Classification Answer Key, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Reactions Classification Answer Key
 - Chapter 2: Essential Elements of Reactions Classification Answer Key
 - Chapter 3: Reactions Classification Answer Key in Everyday Life
 - Chapter 4: Reactions Classification Answer Key in Specific Contexts
 - ∘ Chapter 5: Conclusion
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