

TUTORIAL 3

ALKENES & CYCLOALKENES

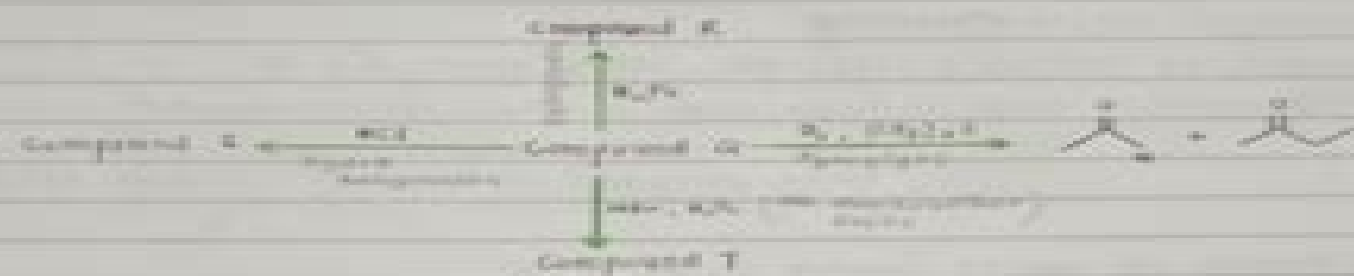
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Q1

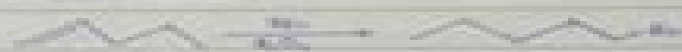
Alkene can undergo various reactions. Identify the structure of R, S, T and U.



Compound R (alkene)



Compound T (anti-Markovnikov addition)



Compound S (hydrobromide)



Compound U (hydrobromide)



Q2

a) Define hydrocarbon compounds

A class of organic compounds made up of only the elements carbon (C) and hydrogen (H)

b) Indicate the difference between propene and propane

Propene	Propane
- Propene is a saturated hydrocarbon compound	- Propane is a unsaturated hydrocarbon compound
- carbon-carbon double bond	- carbon-carbon single bond in the structure
- fully saturated to bond	- double bond and single bond

Reaction Lab Answer Key Hydrocarbon

R Sanford

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Reaction Lab Answer Key Hydrocarbon:

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