

Reaction Energy

Reaction Quiz: Thermodynamics

In the space provided, write the letter of the best answer for each multiple-choice question or the appropriate calculation.

- _____ 1. The sign convention for the study of thermodynamics is that
 - a. energy + changes during heating.
 - b. energy + changes during + bonds dissociation.
 - c. a rise - changes during chemical reactions.
 - d. heat is work.
- _____ 2. Energy is measured in units of
 - a. joules.
 - b. calories.
 - c. grams.
 - d. grams.
- _____ 3. Heat gained is used to measure enthalpy?
 - a. H.
 - b. ΔH .
 - c. ΔH .
 - d. None of the above.
- _____ 4. The specific heat of a substance is the amount of energy needed to

raise the temperature of

 - a. 1 g by 1 K.
 - b. 1 mol by 1 g.
 - c. 1 g by 1 K.
 - d. 1 mol by 1 K.
- _____ 5. Effects of the following on the heat of combustion (the energy) released to

increase the temperature of a substance?

 - a. increase of the substance.
 - b. the specific heat of the substance.
 - c. the mass of the substance.
 - d. the temperature of the substance.

Reaction Energy Section Quiz Modern Chemistry

Lingsheng Yao



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