Student Name:	Print.	Date
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Periodic Table Puzzle

						1	

Using the following hat of class, place each element in its proper position on the periodic table. The letters bear no similarity to the octual symbols for the elements. The trumpated periodic table above these only the first four periods of the representative elements and the nable gases. Hint - place each group in its proper position first, and then arrange the elements in each group.

CLUES

- The following elements are grouped together into faculties: ERH, GSI, NLW, FQK, CDAB, UFZ, YXJ, and MYOT.
- C is a more reactive metal than A and B.
- O forms an oxide with oxygen in the following manner: Ox0s.
- 4. If is used as a medication for manic depression (bipolar disorder).
- Thus the highest mass on the periodic table.
- 6. Zisa liquid.
- I is a non-mutal that was originally entracted from uring.
- 8. If has three valence electrons on the third energy level and can be found in the kitchen.
- 9. G has a higher electron affinity than both Y and K.
- 10. S is a semi-metal that is used in the manufacturing of computer chips.
- 11. It is the largest atom of the periodic table.
- 12. N has a smaller I" ionization energy than P or X.
- 13. V does not have an electronogativity value.
- 14. Y is more dense than I and X
- 15. Q has a senaller ionization energy than E.
- 16. Whas a greater effective nuclear charge than L but less than N.
- 17. Hi reacts with water to form a white, milky base solution and hydrogen gas.
- 18. A is an element which can form both an axion and cation.
- S loses all of its 3" shell electrons to form a +4 charge.
- 20. P is the most electrosegative element on the periodic table.
- 21. O is lighter than air and has the lowest moiting point of all the elements.
- 22. U has a more negotive electron affinity than both M and W.
- 23. R is larger than E but smaller than H.
- 24. G is named after a country in Europe.
- 25. N is the only member of its group that is a gas.

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- 1. Understanding the eBook Periodic Trends Puzzle Answer Key
 - The Rise of Digital Reading Periodic Trends Puzzle Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Periodic Trends Puzzle Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Puzzle Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Periodic Trends Puzzle Answer Key
 - Personalized Recommendations
 - Periodic Trends Puzzle Answer Key User Reviews and Ratings
 - Periodic Trends Puzzle Answer Key and Bestseller Lists
- 5. Accessing Periodic Trends Puzzle Answer Key Free and Paid eBooks
 - Periodic Trends Puzzle Answer Key Public Domain eBooks
 - Periodic Trends Puzzle Answer Key eBook Subscription Services
 - Periodic Trends Puzzle Answer Key Budget-Friendly Options
- 6. Navigating Periodic Trends Puzzle Answer Key eBook Formats

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- Periodic Trends Puzzle Answer Key Compatibility with Devices
- Periodic Trends Puzzle Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Puzzle Answer Key
 - Highlighting and Note-Taking Periodic Trends Puzzle Answer Key
 - Interactive Elements Periodic Trends Puzzle Answer Key
- 8. Staying Engaged with Periodic Trends Puzzle Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Puzzle Answer Key
- 9. Balancing eBooks and Physical Books Periodic Trends Puzzle Answer Key
 - Benefits of a Digital Library
 - o Creating a Diverse Reading Collection Periodic Trends Puzzle Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Puzzle Answer Key
 - Setting Reading Goals Periodic Trends Puzzle Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Puzzle Answer Key
 - Fact-Checking eBook Content of Periodic Trends Puzzle Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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