# Lab 1 - Melting Point

#### Part I - Purpose and Procedure

The purpose of the lab is determine the melting point of samples of pure urea, pure cinnamic acid, and varying mixtures of the two.

#### Procedure

#### Melting Points of Pure Urea and Cinnamic Acid

- 1. Crush a sample of both pure urea and cinnamic acid into powder using a metal spatula
- Push the open end of a capillary into the powder in order to force the powder into the capillary, then lightly tap the capillary on a hard surface until the sample is tightly packed at the bottom
- Heat the melting point system to approximately fifteen degrees below the known melting point should be the same for both the urea and the cinnamic acid
- 4. Heat samples at the rate of one degree per minute
- 5. Note the temperature at which the first drops of liquid appear
- 6. Note the temperature at which any trace of solid has disappeared

#### Melting Points of Urea-Cinnamic Acid Mixtures

- 1. Make mixtures of urea and cinnamic acid in the approximate proportions 1:4, 1:1, and 4:1
- 2. Grind mixtures to ensure thoroughly mixed
- 3. Refer to the steps of the previous procedure for the remainder

#### Materials

- Weigh boats
- Optimelt® Automated Melting System
- Capillaries
- Metal spatula

#### Substances

Substance:	Amount	Melting Point		
Urea	<1 g	132_5 ° C - 133 ° C		
Cinnamic Acid	<1 g	132.5 ° C-133 ° C		

# **Organic Chemistry Lab Manual Melting Point Answers**

**Denise Guinn, Rebecca Brewer** 

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#### **Table of Contents Organic Chemistry Lab Manual Melting Point Answers**

- 1. Understanding the eBook Organic Chemistry Lab Manual Melting Point Answers
  - The Rise of Digital Reading Organic Chemistry Lab Manual Melting Point Answers
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Organic Chemistry Lab Manual Melting Point Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chemistry Lab Manual Melting Point Answers
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Organic Chemistry Lab Manual Melting Point Answers
  - Personalized Recommendations

- Organic Chemistry Lab Manual Melting Point Answers User Reviews and Ratings
- Organic Chemistry Lab Manual Melting Point Answers and Bestseller Lists
- 5. Accessing Organic Chemistry Lab Manual Melting Point Answers Free and Paid eBooks
  - Organic Chemistry Lab Manual Melting Point Answers Public Domain eBooks
  - o Organic Chemistry Lab Manual Melting Point Answers eBook Subscription Services
  - o Organic Chemistry Lab Manual Melting Point Answers Budget-Friendly Options
- 6. Navigating Organic Chemistry Lab Manual Melting Point Answers eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Organic Chemistry Lab Manual Melting Point Answers Compatibility with Devices
  - o Organic Chemistry Lab Manual Melting Point Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Organic Chemistry Lab Manual Melting Point Answers
  - Highlighting and Note-Taking Organic Chemistry Lab Manual Melting Point Answers
  - Interactive Elements Organic Chemistry Lab Manual Melting Point Answers
- 8. Staying Engaged with Organic Chemistry Lab Manual Melting Point Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chemistry Lab Manual Melting Point Answers
- 9. Balancing eBooks and Physical Books Organic Chemistry Lab Manual Melting Point Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chemistry Lab Manual Melting Point Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chemistry Lab Manual Melting Point Answers
  - Setting Reading Goals Organic Chemistry Lab Manual Melting Point Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chemistry Lab Manual Melting Point Answers
  - Fact-Checking eBook Content of Organic Chemistry Lab Manual Melting Point Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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