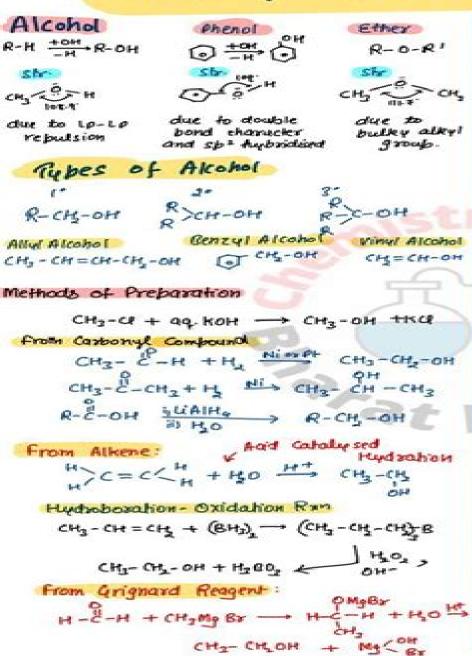
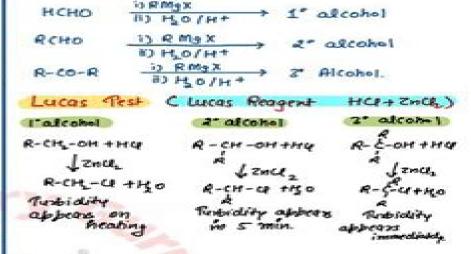
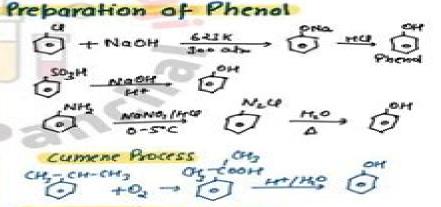
Alcohol, Phenol and Ethers









Physical Properties

- · Due to strong hydrogen bonding b/w molecules of alcohol, they have higher bpt as compand to molecules of alkane having some molecular mass.
- As molecular weight increases and surface area also increases and there is increase in Van der waal force. As a Result m.pt increases.

Organic Chem Phenol Study Guide

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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