

9. Alkene usually show which type of reaction ?

- (a) Substitution (b) **Addition** (c) Elimination (d) Rearrangement

10. when 3-phenyl propene reacts with HBr in the presence of peroxide, the major product form is

- (a) **2-bromo-1-phenyl propane** (b) 1,2-dibromo-3-phenyl propane
(c) 3-(0-bromo phenyl)propane (d) 1-bromo-3-phenyl propane

11. The addition of HBr to pent-2-ene gives

- (a) 2-bromo pentane (b) **3-bromo pentane**
(c) Mixture of(A) and (B) (d) 1- bromopentane

12. Addition of HCl to propene in presence of peroxide gives

- (a)**1-Chloropropane** (b) 2-Chloropropane
(c) 3-Chloropropane (d) Chloropropene peroxide

13. Ethylene dibromide on heating with alc.KOH gives mainly.

- (a) Ethane (b) Ethylene (c) **Acetylene** (d) Ethyl bromide

14. Reduction of acetylene in presence of Ni/Pd gives

- (a) **Ethane** (b) Ethene (c) Ethanol (d) Ethanamine

15. Point out the wrong statement in relation to the structure of Benzene.

- a. **It forms only one monosubstituted derivative**
b. The C-C bond length in benzene is uniformly 1.397 Å
c. It is a resonance hybrid of a number of canonical forms
d. It has three delocalised p- molecular orbitals

16. Which is not aromatic hydrocarbon?

- (a) Benzene (b) Toluene (c) **phenol** (d) Napthalene

17. Benzene reacts with CH_3COCl in presence of AlCl_3 to give

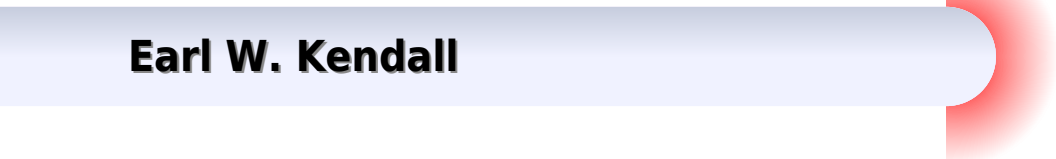
- (a) $\text{C}_6\text{H}_5\text{Cl}$ (b) $\text{C}_6\text{H}_5\text{COCl}$ (c) **$\text{C}_6\text{H}_5\text{COCH}_3$** (d) $\text{C}_6\text{H}_5\text{CH}_3$

18. Nitration of Benzene is

- (a) **Electrophilic Substitution** (b) Nucleophilic Substitution
(c) Electrophilic addition (d) Free radical Substitution

Organic Inorganic Chemistry Fybsc

Earl W. Kendall



Organic Inorganic Chemistry Fybsc:

Organic and Inorganic Chemistry Dakota Rooney, 2021-11-16 Organic and inorganic chemistry are sub disciplines of chemistry that study organic and inorganic compounds respectively Organic chemistry studies the structure properties and reactions of organic compounds Such compounds contain carbon in covalent bonding It is important to study their structure to determine their chemical composition and formula This branch of chemistry studies the physical and chemical properties of organic compounds and evaluates their chemical reactivity to understand their behavior Inorganic chemistry focuses on the synthesis and behavior of inorganic and organometallic compounds Inorganic compounds are derived from nature as minerals This book is a valuable compilation of topics ranging from the basic to the most complex theories and principles in the field of organic and inorganic chemistry Some of the diverse topics covered in this book address the varied branches that fall under this category It will provide comprehensive knowledge to the readers **A Handbook of Organic (inorganic)**

Chemistry William Gregory (Professor of Chemistry in the University of Edinburgh.), 1852 *Organic-Inorganic Chemistry Molecular Student Set #62053* Indigo Instruments, 2003-05-01 *Basic Organic and Inorganic Chemistry* Dakota Rooney, 2025-08-25 Within the subject of chemistry organic and inorganic chemistry constitute distinct sub disciplines each dedicated to the study of organic and inorganic compounds respectively Organic chemistry delves into studying the properties structure and reactions of organic compounds Organic compounds are distinguished by carbon atoms bonded via covalent linkages Understanding the structural nuances is crucial for discerning their chemical composition and formula This branch of chemistry scrutinizes the physical and chemical attributes of organic compounds delving into their reactivity to understand their behaviour Conversely inorganic chemistry is primarily concerned with synthesizing and behavior of inorganic and organometallic compounds Inorganic compounds stemming from natural sources are often sourced as minerals The topics included in this book on organic and inorganic chemistry are of the utmost significance and bound to provide incredible insights to readers Most of the topics introduced herein cover new concepts and applications of organic and inorganic chemistry This book with its detailed coverage will prove immensely beneficial to professionals and students involved in this area at various levels FUNDAMENTALS OF INORGANIC AND ORGANIC CHEMISTRY. ILIA.

MANOLOV, 2023 **Organic and Inorganic Chemistry** R. K. Singh, 2020 *A Handbook of Organic (inorganic) Chemistry* William Gregory (Professor of Chemistry in the University of Edinburgh.), 1852 Valuepack: Organic Chemistry: International Edition/Chemistry Paula Yurkanis Bruice, Catherine E. Housecroft, Edwin Constable, 2007-08-02 This Valuepack consists of Chemistry An Introduction to Organic Inorganic and Physical Chemistry by Housecroft Constable ISBN 9780131275676 Organic Chemistry International Edition 5 e by Bruice ISBN 9780131996311 **Inorganic Chemistry** Rory Reid, 2018-01-11 Inorganic chemistry deals with the synthesis and behavior of inorganic and organometallic compounds This field covers all chemical compounds except the myriad organic compounds which are the subjects of organic chemistry

The distinction between the two disciplines is far from absolute as there is much overlap in the subdiscipline of organometallic chemistry. Today our understanding of chemical bonding, molecular reactivities and various other fundamental chemical problems rests heavily on our knowledge of the detailed behaviour of electrons in atoms and molecules. This book describes in detail some of the basic principles, methods and results of quantum chemistry that lead to our understanding of electron behaviour. The basic aspects of inorganic chemistry are presented significantly in this book. Many applications and practical problems are described. The order of the techniques included is conventional and would be liked by students. The chapters have been arranged in a conventional way as it may be easy for students to pass from one to another chapter with continuity. *Basic Inorganic Chemistry* Earl W. Kendall, 1944

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