

Organic Chemistry 11th Edition Solomons Test Bank

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Package title: Solomons Test Bank
Course Title: Solomons 11e
Chapter Number: 2

Question type: Multiple choice

1) Which of the following compounds contains polar covalent bonds?

- A) CS_2
- B) LiF
- C) F_2
- D) CH_3F
- E) None of these choices.

Answer: D

Topic: Polar Covalent Bonds
Section Reference 1: 2.2
Difficulty: Easy

2) Which molecule does not have a dipole moment?

- A)  A benzene ring with fluorine atoms at the 1 and 2 positions (ortho).
- B)  A benzene ring with fluorine atoms at the 1 and 3 positions (meta).
- C)  A benzene ring with fluorine atoms at the 1 and 4 positions (para).
- D)  A benzene ring with fluorine atoms at the 1, 2, 4, and 5 positions.
- E) None of these choices.

Answer: D

Topic: Molecular geometry, polarity
Section Reference 1: 2.3

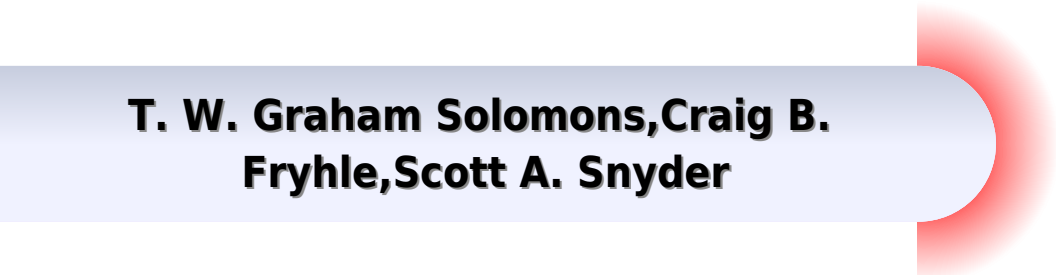
Difficulty: Easy

3) Of the following solvents which one does *not* have a zero dipole moment?

- A) Pentane
- B) Cyclohexane
- C) Diethyl ether

Organic Chemistry Solomons 11 Solution

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**



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