Organic chemistry II (alkanoic acids and alkanols)

- 1. (i) Ethylbutanoate
 - (ii) CH₃CH₂CH₂
 - (iii) Esters
- a) -CH- CH- CH₂ CH CH₂ CH CH₂ CH
 - b) Polypheny/ ethane
- Plastics may contain chlorine or fluorine compounds apart from hydrogen and carbon when burnt, fluorine and chlorine compounds are released into the air destroying Ozone layer
- 4. $(NH_d)_2 CO_{3(\eta)}$ $2NH_{3(g)} + CO_{2(g)} + H_2O_{(l)}$
- The first amount of soap precipitates Ca²⁺ (sag) and Mg²⁺ (sag) ions and soften water.
 Then additional soap dissolves oil from the fabric.
- 6. a) CH₂CH₂ O O CH₂CH₃

$$-NH-CH-C-NH-C-C-C-$$

- Signatura de la companio del companio de la companio della compani
- b) 0.00005mol. P = 0.515 g of monomer.

25

= 1.0 mole of poly mer = <u>IX 0.515</u> = 10300 g

$$RFM (C_sH_sND_s)n = 48 + 9 + 32 = 103$$

= $(C_sH_sNO_s) = 10300$
 $103n = 10300$
 $n = 100 \text{ molecules}$

- Agent A magnesium salt formed is soluble hence doesn't form scum
- 8. (a) Styrene/Phenylethene
 - (b).Addition polymerization
 - (c) can be made into different shapes easily
 - are cheaper
 - are not corroded by acids, alkalis or air
 - are stronger and long lasting
 - are water-proof
- Add water to the mixture and shake where ethanol dissolves in water while pentane is immiscible.
- *MAT

7.

- Transfer the mixture in a separating funnel and allow it to settle when pentane floats on top of water-ethanol mixture.
- *MAT
- Turn on the tap to collect water-ethanol mixture while pentane remains in the separating funnel.

Any 1 correct

- Separate ethanol from water by fractional distillation based on the differences in boiling points.
- (a) Is 100% ethanol/is pure ethanol without water in it

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- (b) 30°C and yeast
- 11. (ii) R = Ar.

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Kaplan Test Prep

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