

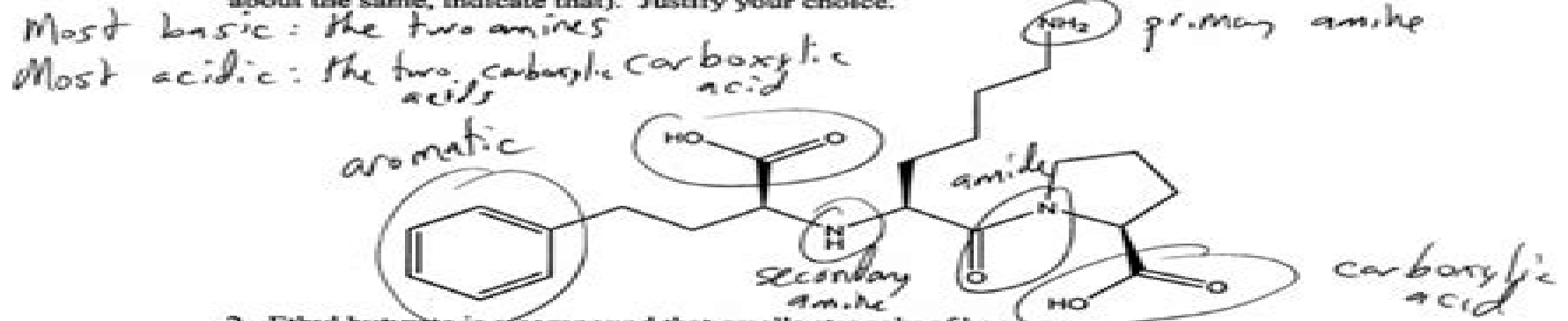
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Exam 1

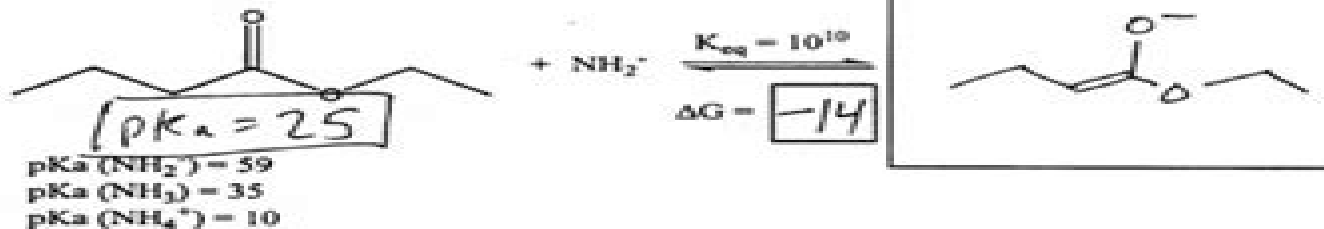
Tuesday, October 27, 2009

Answer as many questions as you can in the time allowed. Please be as clear and concise as possible.

1. Zestril is a bestselling medication used to treat high blood pressure.
 - a. Identify all the functional groups in morphine. Classify as primary, secondary, tertiary where relevant.
 - b. Identify the most acidic site and the most basic site in Zestril (if more than one will be about the same, indicate that). Justify your choice.



2. Ethyl butyrate is a compound that smells strongly of banana.
 - a. Ethyl butyrate is somewhat acidic. Draw its conjugate base.
 - b. Ethyl butyrate can be deprotonated with NH_2^- . The K_{eq} for this reaction is 10^{10} . What is the pK_a of ethyl butyrate? What is the ΔG for this reaction?



Organic Chem 1 Exam Une

A Loxley



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