



Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = 30.0 \text{ g/mol}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H₂O) gmm:

$$(2)(1.0) + (1)(16.0) = 18.0 \text{ g/mol}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = 88.9\%$$

$$\frac{45 \text{ g NO} \times \frac{1 \text{ mol NO}}{30 \text{ g NO}} \times \frac{4 \text{ mol NO}}{4 \text{ mol NO}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times 18 \text{ g H}_2\text{O}}{1} = 6.75 \text{ g H}_2\text{O}$$

Percent Yield Calculations Answers

**Christopher R. Hren, John T.
Moore, Peter J. Mikulecky**



Percent Yield Calculations Answers:

Whispering the Secrets of Language: An Mental Quest through **Percent Yield Calculations Answers**

In a digitally-driven earth wherever displays reign supreme and immediate interaction drowns out the subtleties of language, the profound secrets and mental subtleties concealed within words usually get unheard. Yet, located within the pages of **Percent Yield Calculations Answers** a captivating fictional treasure pulsing with organic feelings, lies an exceptional quest waiting to be undertaken. Penned by an experienced wordsmith, this enchanting opus attracts viewers on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within the cloth of each word. Within the mental depths of this touching review, we can embark upon a heartfelt exploration of the book is core styles, dissect its captivating writing type, and fail to the effective resonance it evokes deep within the recesses of readers hearts.

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