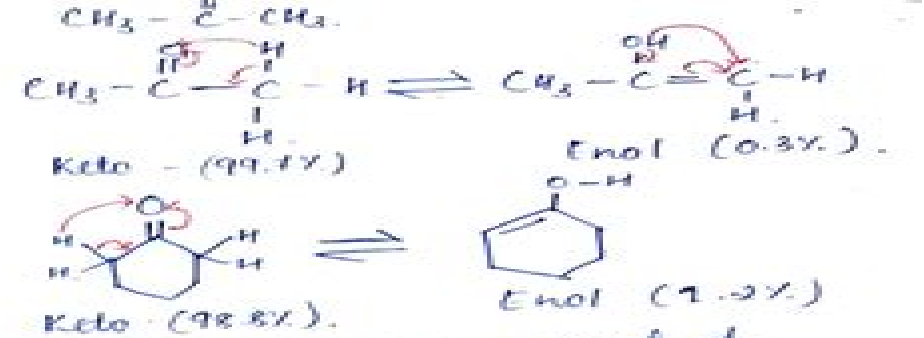
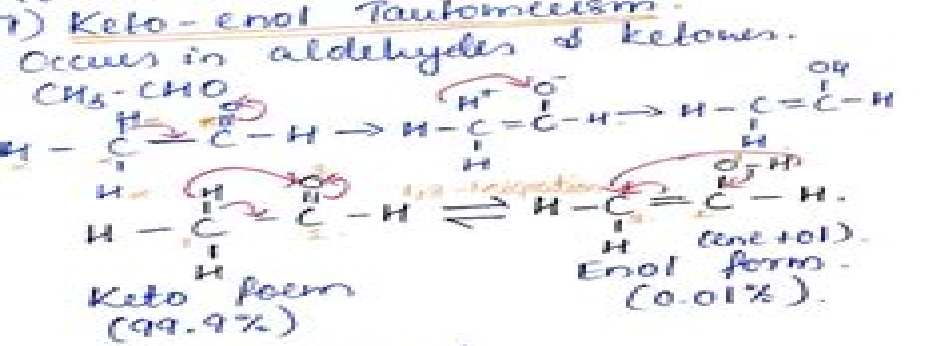


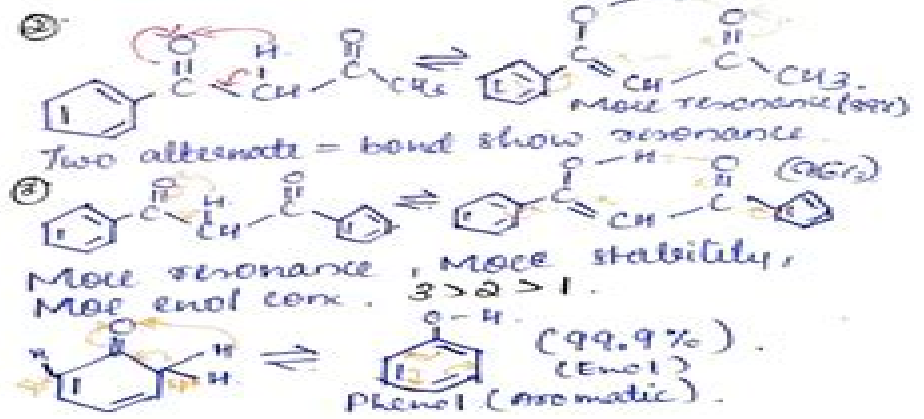


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2. Carboxylic acid - Ester
3. Aldehydes - Ketone - Enones
Unsubstituted alcohol - Unsat. Ethers
4. Diene - Allene - Alkyne
5. Ar. alcohol - Phenol - Ar. Ethers
6. 1°, 2°, 3° Amine
7. Amides - Oximes
8. Nitroalkanes - Alkyl nitrite
9. Cyanides - Isocyanides

Tautomerism
 Same molecular formula, diff. fn. group
 There 2 comp. exist inter-converted
 state, no stable existence.
 They always exist in dynamic eqib.
 It occurs due to migration of H atoms



Conc. of Keto is prominent & enol is negligible.
 C=O bond energy is greater than C=C bond energy.
 (364 kJ/mol) vs (254 kJ/mol)
 CH₃-C(=O)-CH₂-CH₃ ⇌ CH₃-C(OH)=CH-CH₃
 More stable Less stable
 ∴ this H more acidic.
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