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55 multiple	e-choice questio	ns (3 pts each)	+ 6 fill-in ques	tions (40 pts) = Tot	al (205 pts)
1) How ma	ny unpaired el	ectrons are pre	sent in the elect	ron configuration o	f the O ator
A) none	B) one	C) two	D) three	E) four	
2) Which o	f the following	contain(s) pola	ar covalent bon	ds?	
A) H ₂	B) NaI	C) Cl2	D) H ₂ O	E) both A and C	
3's William in	the formal char	re and hubridia	ration for the hi	ghlighted Carbon?	
	B) +1, sp ²			E) 0, sp	H ₃ C-N
400024800000		5190 ms 32	6000000000000000	1001144 FF 310	92001379280
		est describes th	ne hybridization	of C in carbocation	on CH3* ?
A) sp3	B) sp2	C) sp	D)s		
5) what orb	vital is the nitros	en lone pair re	siding in the fo	flowing molecule?.	
A) sp ³ B):					U =
					N
6) Identify	the most acidic	carboxylic ac	id.	. 0	
_ I	_ ĭ		Ľ'Ĺ	ĭ rJ	
F- ~ ~o	H Br →	OH OF	OH P	OH FT E	
Answer is i				0:000	
7) The con	jugate acid of I	I ₂ O is			
A) HO-	B) H ₃ O	C) H ₃ O ⁺	D) H ₂ O	E) H ₂ O ⁺	
ex House me	ny potential ste	manifement and	et for the follow	ing malanda	→ "CH
	2 C) 3 D) 4		at for the follow	ving morecure.	r
., .,		100			₩ CH
			CONTRACTOR WAS INC.		
9) what is t	he name of the		pound? :/x-4-ethy/hexar		1.00

10) which of the following step is part of the propagation in the methane chlorination?

A) CI-CI \longrightarrow 2 CI · B) H_3C · + CI-CI \longrightarrow H_3C-CI + CI · C) H_3C-H + · CI \longrightarrow H_3C · + H-CI D) B and C

Answer is D

11) Based on the Hammond postulate, which of the following transition state best explains the following reaction step?

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John M. McIntosh

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